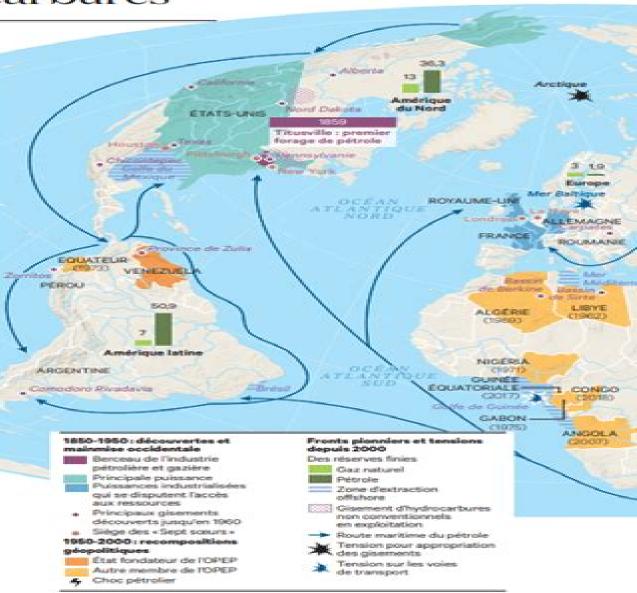
# Les hydrocarbures

#### Un siècle et demi d'exploitation

Des millions d'années de décomposition de matière organique ont donné naissance aux hydrocarbures, piégés dans des réservoirs naturels souterrains. Dès l'Antiquité, ces hydrocarbures découverts de facon fortuite dans des nappes superficielles ont été exploités localement, 1859 est cependant l'année de naissance du pétrole moderne: le 27 août, un Américain fore le premier puits en Pennsylvanie. Ce premier forage ouvre la voie d'une utilisation massive de l'or noir. Une industrie se met en place. associant extraction, transport. et raffinage, entre les mains de grandes compagnies. Progressivement, le marché s'étend au reste du monde. Le décalage entre les sites de production et les lieux de consommation devient un enjeu majeur pour les puissances européennes qui cherchent à sécuriser leur approvisionnement. Leur concurrence pour l'appropriation des ressources est particulièrement forte au Moven-Orient, Longtemps marginalisée en raison notamment des contraintes liées au transport, la production de gaz se généralise dans la seconde moitié du XX<sup>a</sup> siècle. Les avantages qu'il présente par rapport au pétrole (moins d'émissions de CO<sub>2</sub>, répartition des ressources) en font une source d'énergie très demandée, Cependant, le caractère fini des réserves, la hausse de la demande et la recherche d'une sécurité énergétique, conduisent certains pays à se tourner vers de nouveaux fronts émergétiques.



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**Kathleen Armour** 

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web jan 29 2020 step 3 determine the number of bonds in the molecule covalent bonds are formed when one electron from each atom forms an electron pair step 2 tells how many electrons are needed and step 1 is how many electrons you have subtracting the number in step 1 from the number in step 2 gives you the number of electrons needed to complete

 $\textbf{vsepr using lewis structures and vsepr to determine molecular geometry} \cdot \texttt{May } 12\ 2023$ 

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web this effect determines the atom s geometry and bond angles two regions will be 180 apart three regions will be 120 apart and four regions will be 109 5 apart geometry determination a determine the lewis dot structure of the molecule or ion

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