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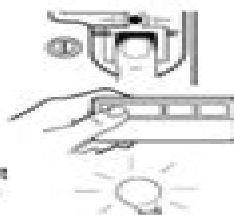
Models 373W 315 MHz Security+® 3-Button Remote Control
OWNER'S MANUAL

Only for use with 315 MHz Security+® garage door openers and universal receivers that have a purple "learn" button.

Instructions are Described and Illustrated Below



1. Press and release the "learn" button on the motor unit. The learn indicator light will glow steadily for 30 seconds.
2. Within 30 seconds, press and hold the button on the hand-held remote.
3. Release the button when the motor unit light blinks. It has learned the code. If light bulbs are not installed, two clicks will be heard.



To Erase All Codes From Motor Unit Memory

To deactivate any unwanted remote, first erase all codes:

Press and hold the "learn" button on motor unit until the learn indicator light goes out (approximately 6 seconds). All previous codes are now erased. Reprogram each remote or keyless entry you wish to use.



To Control the Opener Lights

NOTE: Not available with universal receiver.

With 315 MHz Security+® remote control, a remote push button can be programmed to operate the opener lights without opening the door.

1. With the door closed, press and hold the remote button that you want to control the light.
2. Press and hold the Light button on the door control.
3. Press and hold the Lock button on the door control.
4. After the opener lights flash, release all buttons.

Test by pressing the remote push button. The opener lights should turn on or off but the door should not move.

FOR SERVICE DIAL OUR TOLL FREE NUMBER:
US: 1-800-528-2817
CANADA: 1-800-654-4736

⚠ WARNING

To prevent possible **SERIOUS INJURY** or **DEATH** from a moving gate or garage door:

- ALWAYS keep remote controls out of reach of children. NEVER permit children to operate, or play with remote control transmitters.
- Activate gate or door **ONLY** when it can be seen clearly, is properly adjusted, and there are no obstructions to door travel.
- ALWAYS keep gate or garage door in sight until completely closed. NEVER permit anyone to cross path of moving gate or door.

Open/Close/Stop Operation

NOTE: Not available with universal receiver.

Your 315 MHz Security+® remote control can be programmed to operate one door using all 3 buttons: the large button will only open the door, the middle button will close the door, and the third button will stop the door's movement. You may set up this feature as follows:

1. With the door closed, press and hold the large remote push button.
2. Press and hold the Lock button on the door control.
3. Press and hold the door control push bar.

When the opener lights flash, release all buttons. Test by pressing the large (Open) button on the remote. The door should open. Press it again while the door is open and nothing should happen. Press the middle (Close) button while the door is moving and it should close. Press the third (Stop) button and the door should stop immediately.



The Remote Control Batteries

⚠ WARNING

To prevent possible **SERIOUS INJURY** or **DEATH**:

- NEVER allow small children near batteries.
- If battery is swallowed, immediately notify doctor.

The lithium batteries should produce power for up to 5 years. To replace, pry open case with visor clip or screwdriver, as shown. Insert batteries positive side up (+).

Dispose of old batteries properly.



NOTICE: To comply with FCC and/or Industry Canada (IC) rules, adjustment or modifications of this receiver and/or transmitter are prohibited, except for changing the code setting or replacing the battery. THERE ARE NO OTHER USER SERVICEABLE PARTS.

Tested to Comply with FCC Standards FOR HOME OR OFFICE USE. Operation is subject to the following two conditions: (1) this device may not cause harmful interference, and (2) this device must accept any interference received, including interference that may cause undesired operation.

Replacement Parts

3V2032 Lithium battery:10A20
Visor clip:29B137

Liftmaster 3 Hp Manual

S Ashworth



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Air Force Manual United States. Department of the Air Force,1964 Sweet's General Building & Renovation ,1999

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound s normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg

experiment 11 vapour pressure proctech 2ce3 lab - Jul 07 2022

the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine

the vapour pressure of a pure liquid at various temperatures

[exploring boiling points sharmin khatun chem 1141 studocu](#) - May 17 2023

exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

boiling point determination lab report paperap com - Sep 09 2022

dec 7 2019 1987 the sample essay on boiling point determination lab report deals with a framework of research based facts approaches and arguments concerning this theme to see the essay's introduction body paragraphs and conclusion read on should be based on observations made during the experiment

boiling point lab report boiling point refractive index and - Aug 20 2023

laboratory partners samantha and mackenzie goals ☐ correctly determine the boiling point of an unknown liquid ☐ measure the refractive index of an unknown ☐ use the gas chromatogram to determine the mass percentage of a mixture and

boiling point determination faculty of science - Jun 18 2023

50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

chem1212 boiling point lab report course hero - Aug 08 2022

introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I aim to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a Bunsen burner boiling

[6.2b step by step procedures for boiling point determination](#) - Mar 15 2023

there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a Thiele tube the most straightforward method uses a Thiele tube and has the advantage of using less than 0.5 mL of material