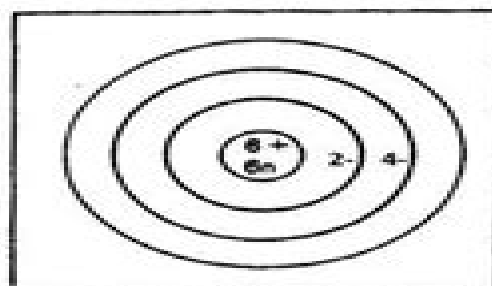
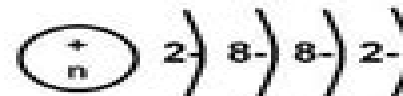


MAKING ATOM MODELS

Draw models of the atoms of the elements indicated in the squares below. The mass numbers and atomic numbers are given for each element. Represent protons with a + sign, neutrons with an n, and electrons with small circles on the rings that represent the energy levels. The model of a carbon atom has been completed for you as an example. Use only the energy levels that are necessary.

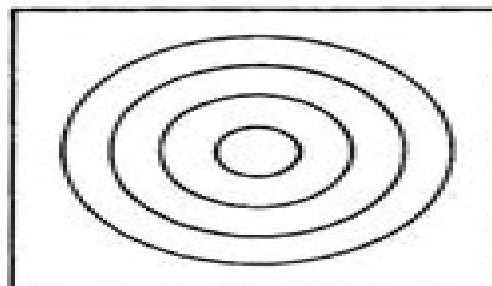
You will need to check the dot diagrams using the picture link



CARBON

Atomic number = 6
Mass number = 12

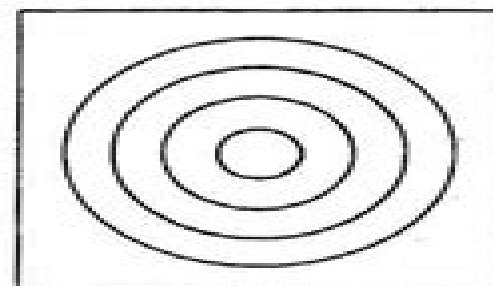
Protons _____
Neutrons _____
Electrons _____
Family _____
Period _____



NEON

Atomic number = 10
Mass number = 20

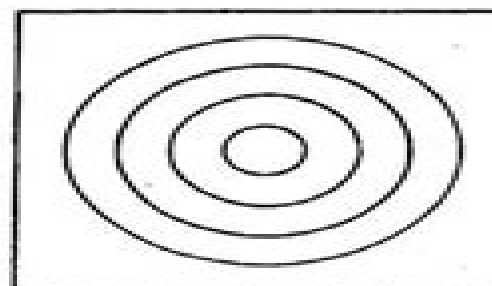
Protons _____
Neutrons _____
Electrons _____
Family _____
Period _____



SODIUM

Atomic number = 11
Mass number = 23

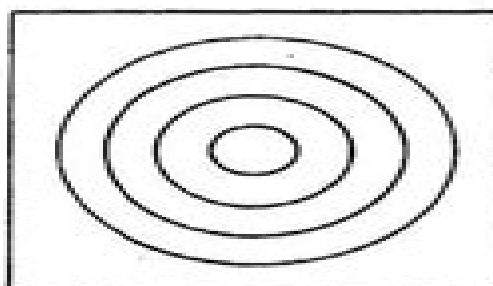
Protons _____
Neutrons _____
Electrons _____
Family _____
Period _____



CHLORINE

Atomic number = 17
Mass number = 35

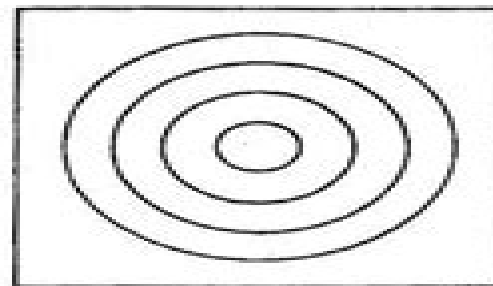
Protons _____
Neutrons _____
Electrons _____
Family _____
Period _____



OXYGEN

Atomic number = 8
Mass number = 16

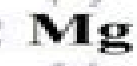
Protons _____
Neutrons _____
Electrons _____
Family _____
Period _____



MAGNESIUM

Atomic number = 12
Mass number = 24

Protons _____
Neutrons _____
Electrons _____
Family _____
Period _____



Making Atomic Models Answer Key

David Kirk



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