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Lab Periodic Trends Worksheet

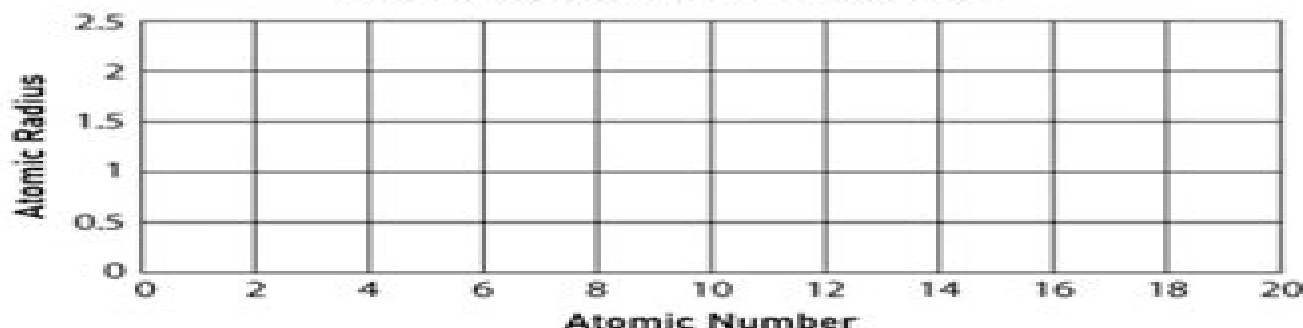
The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

Element	H	He	Li	Be	B	C	N	O	F
Atomic Number	1	2	3	4	5	6	7	8	9
Atomic Radius (Å)	0.79	0.49	2.05	1.4	1.17	0.91	0.75	0.65	0.57
Ionic Radius (Å)	0.012	-	0.76	0.35	0.23	0.16	0.13	1.4	1.33

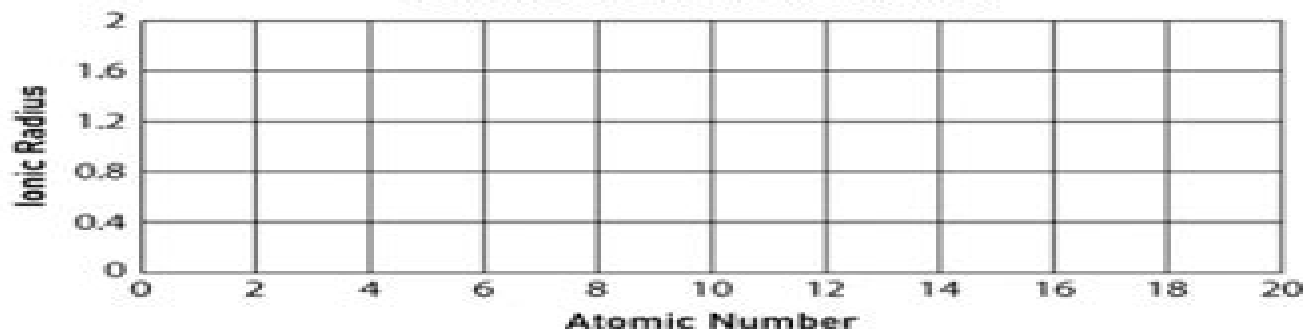
Ne	Na	Mg	Al	Si	P	S	Cl	Ar
10	11	12	13	14	15	16	17	18
0.38	2.23	1.72	1.82	1.46	1.23	1.09	0.97	0.71
-	1.02	0.72	0.535	0.4	0.38	0.37	1.81	-

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.

Atomic Radius vs. Atomic Number



Ionic Radius vs. Atomic Number



Lab Periodic Trends Key

Kathy L. Tyner

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Laboratory Exercises for Preparatory Chemistry Kathy Dodds Tyner, 1994-06 Laboratory Exercises for Preparatory Chemistry is the perfect complement to a one semester preparatory chemistry laboratory course Tyner's manual emphasizes the application of chemistry and the principles of science to everyday life The labs are directly applicable to the real world and often contain supplemental assignments that illustrate an application *Hands-On General Science Activities With Real-Life Applications* Pam Walker, Elaine Wood, 2008-04-21

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Table of Contents Lab Periodic Trends Key

1. Understanding the eBook Lab Periodic Trends Key
 - The Rise of Digital Reading Lab Periodic Trends Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Lab Periodic Trends Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lab Periodic Trends Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lab Periodic Trends Key
 - Personalized Recommendations
 - Lab Periodic Trends Key User Reviews and Ratings

- Lab Periodic Trends Key and Bestseller Lists
- 5. Accessing Lab Periodic Trends Key Free and Paid eBooks
 - Lab Periodic Trends Key Public Domain eBooks
 - Lab Periodic Trends Key eBook Subscription Services
 - Lab Periodic Trends Key Budget-Friendly Options
- 6. Navigating Lab Periodic Trends Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Lab Periodic Trends Key Compatibility with Devices
 - Lab Periodic Trends Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lab Periodic Trends Key
 - Highlighting and Note-Taking Lab Periodic Trends Key
 - Interactive Elements Lab Periodic Trends Key
- 8. Staying Engaged with Lab Periodic Trends Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lab Periodic Trends Key
- 9. Balancing eBooks and Physical Books Lab Periodic Trends Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lab Periodic Trends Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Lab Periodic Trends Key
 - Setting Reading Goals Lab Periodic Trends Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Lab Periodic Trends Key
 - Fact-Checking eBook Content of Lab Periodic Trends Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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