

The  $Q$  value of a nuclear reaction  $A + b \rightarrow C + d$  is defined by

$$Q = [m_A + m_b - m_C - m_d]c^2$$

where the masses refer to the respective nuclei. Determine from the given data the  $Q$ -value of the following reactions and state whether the reactions are exothermic or endothermic.



Atomic masses are given to be

$$m({}_1^2\text{H}) = 2.014102 \text{ u}$$

$$m({}_1^3\text{H}) = 3.016049 \text{ u}$$

$$m({}_6^{12}\text{C}) = 12.000000 \text{ u}$$

$$m({}_{10}^{20}\text{Ne}) = 19.992439 \text{ u}$$

Suppose, we think of fission of a  ${}_{26}^{56}\text{Fe}$  nucleus into two equal fragments,  ${}_{13}^{28}\text{Al}$ . Is the fission energetically possible? Argue by working out  $Q$  of the process. Given  $m({}_{26}^{56}\text{Fe}) = 55.93494 \text{ u}$  and  $m({}_{13}^{28}\text{Al}) = 27.98191 \text{ u}$ .

The fission properties of  ${}_{94}^{239}\text{Pu}$  are very similar to those of  ${}_{92}^{235}\text{U}$ . The average energy released per fission is 180 MeV. How much energy

# M C Q Of Nuclear Chemistry

**Maheshwar Sharon, Madhuri Sharon**



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