

## 8.2

# THE NATURE OF COVALENT BONDING

# KEY

## Section Review

### Objectives

- State a rule that usually tells how many electrons are shared to form a covalent bond
- Describe how electron dot formulas are used
- Predict when two atoms are likely to be joined by a double or a triple covalent bond
- Distinguish between a single covalent bond and other covalent bonds
- Describe how the strength of a covalent bond is related to its bond dissociation energy
- Describe how resonance structures explain bonding
- Identify some exceptions to the octet rule

### Vocabulary

- single covalent bond
- structural formulas
- unshared pairs
- double covalent bonds
- triple covalent bonds
- coordinate covalent bond
- polyatomic ion
- bond dissociation energy
- resonance structures

### Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

When atoms share electrons to gain the 1 configuration of a noble gas, the bonds formed are 2. A 3 pair of valence electrons constitutes a 4 covalent bond. Pairs of valence electrons that are not shared between atoms are called 5. Sometimes two or three pairs of electrons may be shared to give 6 covalent bonds. In some cases, only one of the atoms in a bond provides the pair of bonding electrons; this is a 7 8. 9 is required to break covalent bonds between atoms. The total energy required to break the bond between two covalently bonded atoms is known as the 9.

When it is possible to write two or more valid electron dot formulas for a molecule or ion, each formula is referred to as a 10.

1. stable
2. covalent
3. shared
4. single
5. unshared pairs
6. double / triple
7. coordinate covalent bond
8. energy
9. bond dissociation energy
10. resonance structures

# Nature Of Covalent Bonding Section Review

**Daniel Israel Arnon, L. R. Blinks, Alden  
Springer Crafts, Leonard Machlis**

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*Study Guide to accompany Basic Concepts of Chemistry, 7th Edition* Leo J. Malone, 2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone s Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry

**Advances in Structural Adhesive Bonding** David A. Dillard, 2023-06-10 Advances in Structural Adhesive Bonding Second Edition reviews developments in adhesive bonding for a range of advanced structural engineering applications This new edition has been fully revised to include the latest advances in materials testing and modeling methods lifecycle considerations and industrial implementation Sections review advances in commonly used groups of structural adhesives covering epoxy acrylic anaerobic and cyanoacrylate polyurethane and silicone adhesives along with toughening Other chapters cover various types of adherends and pre treatment methods for structural materials including metals plastics composites wood and joint design and testing including topics such as fracture mechanics life prediction techniques and advanced testing methods This is a valuable guide for all those working with structural adhesives

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**Nanofabrication Towards Biomedical Applications** Challa S. S. R. Kumar, Josef Hormes, Carola Leuschner, 2006-03-06 This book focuses on the materials synthetic methods tools and techniques being developed in the nanoregime towards the life sciences in particular biology biotechnology and medicine Readers from materials science engineering chemistry biology and medical backgrounds will find detailed accounts of the design and synthesis of nanomaterials and the tools and techniques involved in their production for applications in biology biotechnology and medicine

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples

*Commercial Fisheries Abstracts*, 1972

**Mechanical and Thermophysical Properties of Polymer Liquid Crystals** Witold Brostow, 2013-11-27 may never overcome the effects of hysteresis and stress see Chapters 6 and 12 The first sentence of the reference work Handbook of Liquid Crystals reads The terms liquid crystals crystalline liquid mesophase and mesomorphous state are used synonymously to describe a state of aggregation that exhibits a molecular order in a size range similar to that of a crystal but acts more or less as a viscous liquid 2 In other words molecules within a liquid crystalline phase possess some orientational order and lack positional order furthermore the shape of a liquid crystalline sample is determined by the vessel in which it is contained rather than by the orientational order of its aggregated molecules The authors recognized the limitations and imprecision of this definition but like others preceding them could not devise a simple and generally applicable one that is better Regardless the terms liquid crystal and mesophase should not be used interchangeably As mentioned above all liquid crystals are mesophases but all mesophases are not liquid crystals Recent studies employing elaborate and sophisticated analytical techniques have permitted finer distinctions between classical crystals and mesophases At the same time they have made definitions like that from the Handbook of Liquid Crystals somewhat obsolete for reasons other than terminology One part of the problem arises from the use of a combination of bulk properties like flow and microscopic properties like molecular ordering within the same definition

*Telegraphic Journal and Monthly Illustrated Review of Electrical Science*, 1924

**Modern Chemistry** Nicholas D. Tzimopoulos, 1993

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**Annual Review of Plant Physiology** Daniel Israel Arnon,L. R. Blinks,Alden Springer Crafts,Leonard Machlis,1961  
Technical Book Review ,1963      **Introduction to Nonlinear Optical Effects in Molecules and Polymers** Paras N. Prasad,David J. Williams,1991-01-16 Molecular Dynamics in Restricted Geometries Edited by Joseph Klafter and J M Drake  
This investigation of the chemistry and physics of complex systems focuses on the role of spatial restrictions on molecular movement A practical source book for researchers in chemical physics chemical engineering and condensed matter physics and for graduate students in these fields it covers a broad range of topics and critically evaluates methods as they are employed Among the many topics it covers are relaxation and diffusion in restricted geometries excitation energy transfer and photoinduced electron transfer phenomena in some confined systems electron excitation transport in micelles polymers and multilayers and electron excitation transport on polymer chains 1989 0 471 60176 4 437 pp      **Pharmacological Reviews** ,1949

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