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**Ian Pickup**



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web the vapor pressure of a liquid is defined as the pressure exerted by the molecules that escapes from the liquid to form a separate vapor phase above the liquid surface the pressure exerted by the vapor phase is called the vapor or saturation pressure vapor or saturation pressure depends on temperature

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web sep 21 2022 a vapor pressure curve is a graph of vapor pressure as a function of temperature to find the normal boiling point of liquid a horizontal line is drawn from the y axis at a pressure equal to standard pressure

*7 2 vapor pressure chemistry libretexts* - Nov 28 2022

web the graph of the vapor pressure of water versus temperature in figure page index 3 indicates that the vapor pressure of water is 68 kpa at about 90 c thus at about 90 c the vapor pressure of water will equal the atmospheric pressure in

**droplet evaporation modeling by the distillation curve model** - Dec 30 2022

web nov 1 2003 droplet evaporation modeling by the distillation curve model accounting for kerosene fuel and elevated pressures

*kerosene cameo chemicals noaa* - Mar 01 2023

web vapor pressure 0.1 psi at 100 f ntp 1992 vapor density relative to air 4.5 ntp 1992 heavier than air will sink specific gravity 0.8 at 59 f uscg 1999 less dense than water will float

*vapor pressure as a function of temperature the trend lines are* - Jun 04 2023

web in fact when the temperature increases from 20 to 60 c the vapor pressure of kerosene rises from 4 to 18 mbar 34 this suggests that kerosene cannot be entirely vaporized at 60 c

**icsc 0663 kerosene international programme on chemical** - Jan 31 2023

web kerosene icsc 0663 physical chemical information physical state appearance low viscosity liquid with characteristic odour physical dangers as a result of flow agitation etc electrostatic charges can be generated vapour pressure kpa at 37.8 c 13.7 exposure health effects