

CHEMISTRY PAPER 1

(THEORY)

Maximum Marks: 70

Time Allowed: Three Hours

*(Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.)*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section – A consists of one question having sub-parts of one mark each.

Section – B consists of ten questions of two marks each.

Section – C consists of seven questions of three marks each, and

Section – D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant $R = 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1} = 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}$

$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J}$. $1 \text{ Faraday} = 96500 \text{ coulombs}$.

Avogadro's number = 6.023×10^{23} .

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the brackets: [4×1]

[decreases, CN^- ion, activation energy, catalyst, two, Fe^{2+} ion, carbon, lattice energy, enzyme, five, double, halogen, triple, increases]

- (i) In the Haber process, iron changes the _____ of reaction while molybdenum increases the efficiency of the _____.
- (ii) The number of ions that will be produced when potassium ferrocyanide, $\text{K}_4[\text{Fe}(\text{CN})_6]$, dissolves in water is _____. This shows that _____ is the ligand in the coordination compound.

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