

## Unit

## 12

## Hydrocarbons

## Long Answer Questions

**Q.1** What are hydrocarbons? Give their different classes depending upon their structure.

**Ans. Hydrocarbons**

Hydrocarbons are those compounds which are made up of only carbon and hydrogen elements.

Hydrocarbons are regarded as the parent organic compounds since other organic compounds are considered to be derived from them by replacement of one or more hydrogen atoms by other atoms or group of atoms.

**Classification of Hydrocarbons**

On the basis of structure, hydrocarbons are divided into two main classes:

**(i) Open chain or Aliphatic hydrocarbons**

These are compounds in which the first and the last carbon are not directly joined to each other. The open chains of carbon may be straight or branched.

**Examples;**



Straight chain (n-butane)



Branched chain (isobutane)

**Types of open chain hydrocarbons**

Open chain hydrocarbons have been further subdivided into saturated and unsaturated hydrocarbons.

**(a) Saturated hydrocarbons**

The compounds in which all the four valencies of carbon atoms are fully satisfied (saturated) by single bonds with other carbon atoms and hydrogen atoms are called saturated hydrocarbons. Saturated hydrocarbons are also called alkanes.

Thus, an alkane is a hydrocarbon in which the carbon atoms are connected by only single covalent bond (There are no double or triple covalent bonds in alkanes).

**Examples:**

Methane ( $\text{CH}_4$ ), ethane ( $\text{C}_2\text{H}_6$ ), propane ( $\text{C}_3\text{H}_8$ ) and butane ( $\text{C}_4\text{H}_{10}$ ) are all saturated

# Mcqs Of Hydrocarbons

**Arshad Iqbal**



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## **Table of Contents Mcqs Of Hydrocarbons**

1. Understanding the eBook Mcqs Of Hydrocarbons
  - The Rise of Digital Reading Mcqs Of Hydrocarbons
  - Advantages of eBooks Over Traditional Books
2. Identifying Mcqs Of Hydrocarbons
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mcqs Of Hydrocarbons
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mcqs Of Hydrocarbons
  - Personalized Recommendations
  - Mcqs Of Hydrocarbons User Reviews and Ratings
  - Mcqs Of Hydrocarbons and Bestseller Lists
5. Accessing Mcqs Of Hydrocarbons Free and Paid eBooks

- Mcqs Of Hydrocarbons Public Domain eBooks
- Mcqs Of Hydrocarbons eBook Subscription Services
- Mcqs Of Hydrocarbons Budget-Friendly Options
- 6. Navigating Mcqs Of Hydrocarbons eBook Formats
  - ePub, PDF, MOBI, and More
  - Mcqs Of Hydrocarbons Compatibility with Devices
  - Mcqs Of Hydrocarbons Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mcqs Of Hydrocarbons
  - Highlighting and Note-Taking Mcqs Of Hydrocarbons
  - Interactive Elements Mcqs Of Hydrocarbons
- 8. Staying Engaged with Mcqs Of Hydrocarbons
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mcqs Of Hydrocarbons
- 9. Balancing eBooks and Physical Books Mcqs Of Hydrocarbons
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mcqs Of Hydrocarbons
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Mcqs Of Hydrocarbons
  - Setting Reading Goals Mcqs Of Hydrocarbons
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcqs Of Hydrocarbons
  - Fact-Checking eBook Content of Mcqs Of Hydrocarbons
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
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