

## Analytical Chemistry MCQ Test. September'21.

1. The metal hydroxide which is insoluble in caustic soda but soluble in excess of ammonium hydroxide.
1.  $\text{Cu}(\text{OH})_2$                       2.  $\text{Zn}(\text{OH})_2$                       3.  $\text{Fe}(\text{OH})_2$                       4. None
2. The salt solution which does not react with ammonium hydroxide is -----
1. Calcium nitrate                      2. Zinc nitrate
3. Lead nitrate                      4. Copper nitrate
3. The metal oxide which can react with acids as well as alkali is-----.
1. Silver oxide                      2. Copper (II) oxide
3. Aluminum oxide                      4. Calcium oxide
4. Al metal shows amphoteric behaviour. it can be shown by
1. Its reaction with dil.  $\text{NaOH}$  only.
2. Its reaction with dil.  $\text{HCl}$  only.
3. Its reaction with both dil.  $\text{HCl}$  and dil.  $\text{NaOH}$  .
4. It does not react with either dil.  $\text{HCl}$  or dil.  $\text{NaOH}$

### QUESTION 5

Which of the following hydroxides soluble in excess of  $\text{NaOH}$ ?

1.  $\text{Cu}(\text{OH})_2$                       2.  $\text{Zn}(\text{OH})_2$                       3.  $\text{Fe}(\text{OH})_2$                       4.  $\text{Fe}(\text{OH})_3$
6. The salt which gives a dirty green precipitate with sodium hydroxide solution is.
1. Iron (III) sulphate                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate
7. The salt which gives a reddish-brown precipitate with sodium hydroxide solution is.
1. Iron (III) sulphate                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate
8. The salt which gives a gelatinous white precipitate with Ammonium hydroxide solution is.
1. Iron (III) chloride                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate
9. The salt which gives a chalky white precipitate with Ammonium hydroxide solution is.
1. Lead nitrate                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate

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