# Name the following acids

| 1. HCL   | hydrochloric acid    |
|--|----------------------|
| 2. HClO <sub>4</sub>                             | perchloric acid      |
| <ol> <li>HIO<sub>3</sub></li> </ol>              | iodic acid           |
| 4. HII   | hydroiodic acid      |
| 5. H <sub>2</sub> SO <sub>4</sub>                | sulfuric acid        |
| <ol> <li>H<sub>2</sub>S</li> </ol>               | hydrosulfuric acid   |
| 7. HCN   | hydrocyanic acid     |
| <ol> <li>H<sub>2</sub>CO<sub>3</sub></li> </ol>  | carbonic acid        |
| 9. HBrO <sub>4</sub>                             | perbromic acid       |
| 10.HBrO <sub>3</sub>                             | bromic acid          |
| 11.HC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> | acetic acid          |
| 12.H <sub>3</sub> PO <sub>3</sub>                | phosphorous acid     |
| 13.H <sub>3</sub> P                              | hydrophosphoric acid |
| 14.H <sub>2</sub> CrO <sub>4</sub>               | chromic acid         |
| 15.H <sub>2</sub> CrO <sub>2</sub>               | hypochromous acid    |
| 16.H <sub>2</sub> Cr                             | hydrochromic acid    |

## Give formulas for the following acids

| 17.hydrofluoric acid | HF                             |
|----------------------|--------------------------------|
| 18.nitric acid       | HNO <sub>3</sub>               |
| 19.nitrous acid      | HNO <sub>2</sub>               |
| 20.hydroxic acid     | H <sub>2</sub> O               |
| 21.periodic acid     | HIO <sub>4</sub>               |
| 22. percarbonic acid | H <sub>2</sub> CO <sub>4</sub> |
| 23.hypobromous acid  | HBrO                           |
| 24. bromous acid     | HBrO <sub>2</sub>              |
| 25.permanganic acid  | HMnO <sub>4</sub>              |
| 26.manganic acid     | HMnO <sub>3</sub>              |
| 27. iodous acid      | HIO <sub>2</sub>               |
| 28. sulfurous acid   | H <sub>2</sub> SO <sub>3</sub> |
| 29. sulfuric acid    | H <sub>2</sub> SO <sub>4</sub> |
| 30.perchloric acid   | HCIO <sub>4</sub>              |
| 31.chlorous acid     | HClO <sub>2</sub>              |
| 32.acetic acid       | $HC_2H_2O_2$                   |
|                      | H <sub>3</sub> PO <sub>4</sub> |
| 34. phosphorous acid | H <sub>3</sub> PO <sub>3</sub> |

# **Naming Acids Packet Answers**

**Karin Nielsen-Saines** 

**Naming Acids Packet Answers:** 

Unveiling the Magic of Words: A Review of "Naming Acids Packet Answers"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really aweinspiring. Enter the realm of "Naming Acids Packet Answers," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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