

Name: _____

Lab: Electron Arrangements

Objectives: You will write short-handled electron configurations; draw orbital filling diagrams; and write electron-dot diagrams.

Procedure: Complete the table below.

Element	Atomic Number	Short-handled Electron Configuration	Orbital Filling Diagrams (Use Box Symbols)	Electron-dot Diagram
Lithium	3	$1s^2 2s^1$	$1s^2 \boxed{\uparrow\downarrow}$	$\cdot\cdot$
Boron	5	$1s^2 2s^2 2p^1$		
Carbon	6	$1s^2 2s^2 2p^2$		
Nitrogen	7	$1s^2 2s^2 2p^3$		
Oxygen	8	$1s^2 2s^2 2p^4$		
Fluorine	9	$1s^2 2s^2 2p^5$		
Neon	10	$1s^2 2s^2 2p^6$		
Sodium	11	$1s^2 2s^2 2p^6 3s^1$		
Magnesium	12	$1s^2 2s^2 2p^6 3s^2$		
Aluminum	13	$1s^2 2s^2 2p^6 3s^2 3p^1$		
Silicon	14	$1s^2 2s^2 2p^6 3s^2 3p^2$		

Name Electron Arrangements Ws Answers

J Ma



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MnM_POW-Science-PM-9 (Updated) Neena Sinha, Anita Marwah, MnM_POW Science PM 9 Updated CBSE - ICSE Chemistry Part I Chandan Sengupta, First Publication October 2021 Place of Publication Arabinda Nagar Bankura 722101 This workbook will provide an ample scope in getting exposed to the system of acquiring skills and competence related to the understanding of chemistry It also exposes the student to the concepts of chemistry for enabling the aspirant in acquisition of skills related to chemistry Some of the worksheets are prepared along with model answers Some other worksheets are meant for self assessment and evaluation purposes It is also observed that some of the topics are specific to the referred curriculum Some other topics are varyingly incorporated in other streams of study Culmination of more than two streams will enable the fellow student to cope up with the preparatory works meant for Olympiads and other competitive examinations

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