#### (25) VIDYALANKAR: JEE-MAIN 2014: Paper and Solution

- 63. The metal that cannot be obtained by electrolysis of an aqueous solution of its salts is : 433 Car. 420 Cen CALL AND CHECK CO.
- 6530 (40) Planettenal.
- 64. Resistance of 0.2 M solution of an electrolyte is 50 Ω. The specific conductance of the solution is 1.4 S. m<sup>-1</sup>. The resistance of 0.5 M solution of the same electrolyte is 280 Gl. The molar conductivity of 0.5 M solution of the electrolyte in S m2 mol 2 is :
- 43.5 5 M 102

化工作 医二甲基酚

Chr. S. Sci. Billion.

- 64.7(20)
  - $A_{co} = \frac{k (s/m)}{1000 \times Motarity (mod/L)} = \frac{1}{4 \times 1000 \times 0.5}$ - 4 to 10°4 Sand mod?
- 65. CsCI crystallises in body centred cubic lattice. If "a" is its edge length that subside of the following expressions is convert ?
  - (1)  $r_{C_{1}} + r_{C_{2}} = \frac{\sqrt{3}}{2} \times (2) r_{C_{2}} + r_{C_{2}} = \sqrt{3} \times (3) r_{C_{2}} + r_{C_{2}} = 3 \times (2) r_{C_{2}} + r_{C_{2}} + r_{C_{2}} = 3 \times (2) r_{C_{2}} + r_{C_{2}} + r_{C_{2}} = 3 \times (2) r_{C_{2}} + r_{C_{2}} + r_{C_{2}} + r_{C_{2}} + r_{C_{2}} = 3 \times (2) r_{C_{2}} + r_{C_{2}} +$
- **65%、111%**

Franchista A.

- Consider separate solutions of 0.500 M C<sub>2</sub>H<sub>2</sub>OH(sq), 0.100 M Mg-PO<sub>2</sub>I<sub>2</sub>(sq), 0.250 M KBr(sq) and 0.125 M Na-POJ(ac) at 25°C. Which statement in true about those solutions, assuming all sailts to be strong electrolytes?
  - (1) 0.125 M NavPOutag) has the highest duratic pressure.
  - (2) 0.500 M C-H-OH as the highest own the present.
  - (3) They all have the same esmotic pressure.
  - (4) 0.100 M Mg-(PO<sub>4</sub>)-(aq) has the highest massric pressure.
- 666, (20)

All solutions have same osmoric prisoure.

- 67. In which of the following regitions HiO- acts as a reducing agent?
  - (a) 15-Oc + 251' + 2c' + 261-O
- the Hoth 2e 00s + 201
- (c) 16:05 + 2c' -- 2000" (13) dath, (ch. 40) 9.27 ( bo), 640.
- 6th H<sub>2</sub>O<sub>5</sub> + 2OH<sup>2</sup> + 2e<sup>2</sup> + O<sub>5</sub> + 2H<sub>2</sub>O CNT date (by) 有限的 化皮肤 化成化

- **新学会主要等** 
  - H/O2 O2 211 1-24
  - $H_2O_2 + 2OH\Gamma \rightarrow O_2 + 2H_2O + 2e^2$
- 68. In Sa. 2 reactions, the correct order of reactivity for the following compounds :
  - CILCIC CHIRCHI-CL (CHASCINCI and (CILS)CCI is :
  - (i) CH/CH/CI> CH/CI> (CH/A/CHCI> (CH/A/CEI
  - (2) (CHA-CHCL) CHECHECL) CHECL > (CHA-CCL
  - CO CRECT > (CH deCHCT > CHECHCT > (CH deCH (4) CH/CL> (CH/CH/CL> (CH/A/CHCL> (CH/A/CCL
- 8880 (AD):

Order of reactivity of alicel bulide towards SN-2 reaction is  $CH-X > 1^{\circ} > 2^{\circ} > 3^{\circ}$ 

- 69. The octahedral complex of a metal ion M\* with four monodentate ligands L., L., L. and L. absorb wavelengths in the region of red, green, yellow and blue, respectively. The increasing order of ligand strongth of the four ligands is :
  - (1)  $L_1 \le L_2 \le L_4 \le L_7$  (2)  $L_1 \le L_2 \le L_4 \le L_7$  (3)  $L_4 \le L_7 \le L_7 \le L_7$  (4)  $L_4 \le L_7 \le L_7 \le L_7$

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- 1. Understanding the eBook Jee Main 2014 Paper 2 Solutions For Code K
  - The Rise of Digital Reading Jee Main 2014 Paper 2 Solutions For Code K
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Jee Main 2014 Paper 2 Solutions For Code K
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Jee Main 2014 Paper 2 Solutions For Code K
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Jee Main 2014 Paper 2 Solutions For Code K
  - Personalized Recommendations
  - $\circ$  Jee Main 2014 Paper 2 Solutions For Code K User Reviews and Ratings
  - Jee Main 2014 Paper 2 Solutions For Code K and Bestseller Lists
- 5. Accessing Jee Main 2014 Paper 2 Solutions For Code K Free and Paid eBooks
  - Jee Main 2014 Paper 2 Solutions For Code K Public Domain eBooks
  - Jee Main 2014 Paper 2 Solutions For Code K eBook Subscription Services
  - ∘ Jee Main 2014 Paper 2 Solutions For Code K Budget-Friendly Options

- 6. Navigating Jee Main 2014 Paper 2 Solutions For Code K eBook Formats
  - o ePub, PDF, MOBI, and More
  - Jee Main 2014 Paper 2 Solutions For Code K Compatibility with Devices
  - ∘ Jee Main 2014 Paper 2 Solutions For Code K Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Jee Main 2014 Paper 2 Solutions For Code K
  - Highlighting and Note-Taking Jee Main 2014 Paper 2 Solutions For Code K
  - Interactive Elements Jee Main 2014 Paper 2 Solutions For Code K
- 8. Staying Engaged with Jee Main 2014 Paper 2 Solutions For Code K
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - $\circ\,$  Following Authors and Publishers Jee Main 2014 Paper 2 Solutions For Code K
- 9. Balancing eBooks and Physical Books Jee Main 2014 Paper 2 Solutions For Code K
  - $\circ$  Benefits of a Digital Library
  - Creating a Diverse Reading Collection Jee Main 2014 Paper 2 Solutions For Code K
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Jee Main 2014 Paper 2 Solutions For Code K
  - Setting Reading Goals Jee Main 2014 Paper 2 Solutions For Code K
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jee Main 2014 Paper 2 Solutions For Code K
  - Fact-Checking eBook Content of Jee Main 2014 Paper 2 Solutions For Code K
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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