

CHEMISTRY 10TH - Multiple Choice Questions

→ CHEMICAL EQUILIBRIUM

CHAPTER# 01

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01. Which one of the following statements is false about dynamic equilibrium?
- (a) It takes place in a closed container
 - (b) Concentration of reactant and products are not changed
 - (c) Rate of the forward reaction is equal to the rate of the reverse reaction
 - (d) ✓ Equilibrium cannot be disturbed by any external stress
02. Consider the following reaction and indicate which of the following best describes equilibrium constant expression K_c . $4\text{NH}_3 + 5\text{O}_2 \rightleftharpoons 4\text{NO} + 6\text{H}_2\text{O}$
- (a) ✓ $K_c = \frac{[\text{NO}]^4 [\text{H}_2\text{O}]^6}{[\text{NH}_3]^4 [\text{O}_2]^5}$
 - (b) $K_c = \frac{[\text{NH}_3]^4 [\text{O}_2]^5}{[\text{NO}]^4 [\text{H}_2\text{O}]^6}$
 - (c) $K_c = \frac{[\text{NH}_3][\text{O}_2]}{[\text{H}_2\text{O}][\text{NO}]}$
 - (d) $K_c = \frac{[4\text{NO}][6\text{H}_2\text{O}]}{[4\text{NH}_3][5\text{O}_2]}$
03. A reaction which is never goes to completion is known as reversible reaction. Reversible reaction is represented by:
- (a) dotted lines
 - (b) single arrow
 - (c) ✓ double arrow
 - (d) double straight line
04. When the magnitude of K_c is small, indicates:
- (a) ✓ Reaction mixture contain most of the reactant
 - (b) Reaction mixture contain most of the product
 - (c) Reaction mixture contain almost equal amount of reactant and product
 - (d) Reaction goes to completion
05. For which system does the equilibrium constant, K_c has units of concentration?
- (a) ✓ $\text{N}_{2(g)} + 3\text{H}_{2(g)} \rightleftharpoons 2\text{NH}_{3(g)}$
 - (b) $\text{N}_{2(g)} + \text{O}_{2(g)} \rightleftharpoons 2\text{NO}_{(g)}$
 - (c) $\text{H}_{2(g)} + \text{I}_{2(g)} \rightleftharpoons 2\text{HI}_{(g)}$
 - (d) $\text{CO}_{2(g)} + \text{H}_{2(g)} \rightleftharpoons \text{CO}_{(g)} + \text{H}_2\text{O}_{(l)}$

Multiple Questions On Chemical Equilibrium

G Psacharopoulos



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- Chapter 4: Multiple Questions On Chemical Equilibrium in Specific Contexts
- Chapter 5: Conclusion

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first the chartered building surveyor who identifies breaches and prices their remedies then the chartered valuation surveyor who assesses the impact on the property's reversionary value which is often far less than the price of the

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web 1 to obtain vacant possession 2 to ensure the tenant maintains the property reference the law of property act 1925 section 146 3 rent review time the chapter then looks at the property and fees where it mentions the RICS fee structure which is a long gone system but remembered fondly by many