

13.1

THE NATURE OF GASES

Section Review

Objectives

1. describe the characteristics of ideal gases and real gases
2. distinguish between an ideal gas and a real gas
3. define the macroscopic and kinetic properties of a gas

Vocabulary

- | | | |
|------------------|-------------------------|------------------|
| a. ideal gas | b. pressure | c. pressure |
| d. absolute zero | e. atmospheric pressure | f. absolute zero |
| g. pressure | h. volume | |

Part A Completion

Use the vocabulary terms in parentheses to complete each sentence. Write the word or phrase in the space provided.

1. A gas is a substance that has _____, expanding to fill its container.
2. The _____ of a gas is the force exerted by the gas on the walls of its container.
3. The _____ of a gas is the volume it occupies.
4. The _____ of a gas is the mass of the gas divided by its volume.
5. The _____ of a gas is the temperature of the gas.
6. The _____ of a gas is the pressure exerted by the gas on the walls of its container.
7. The _____ of a gas is the volume it occupies.
8. The _____ of a gas is the mass of the gas divided by its volume.
9. The _____ of a gas is the temperature of the gas.
10. The _____ of a gas is the pressure exerted by the gas on the walls of its container.
11. The _____ of a gas is the volume it occupies.
12. The _____ of a gas is the mass of the gas divided by its volume.
13. The _____ of a gas is the temperature of the gas.
14. The _____ of a gas is the pressure exerted by the gas on the walls of its container.
15. The _____ of a gas is the volume it occupies.
16. The _____ of a gas is the mass of the gas divided by its volume.
17. The _____ of a gas is the temperature of the gas.
18. The _____ of a gas is the pressure exerted by the gas on the walls of its container.
19. The _____ of a gas is the volume it occupies.
20. The _____ of a gas is the mass of the gas divided by its volume.

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Part B True-False

Write **T** for true or **F** for false in the space provided.

1. _____ A gas is a substance that has _____.
2. _____ The _____ of a gas is the force exerted by the gas on the walls of its container.

Nature Of Gases Section Review Answer Key

K Morrison



Nature Of Gases Section Review Answer Key:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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