

Concept Review:

1. Draw the Lewis diagram for each compound.
 - A. a molecule composed of two chlorine atoms
 - B. a molecule composed of a hydrogen atom and a bromine atom

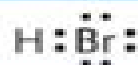
Answers:

Solution

Chlorine has the same valence shell electron configuration as fluorine, so the Lewis diagram for a molecule composed of two chlorine atoms is similar to the one for fluorine:



Bromine has the same valence shell electron configuration as fluorine, so the Lewis diagram for a molecule composed of a hydrogen atom and a bromine atom is similar to that for hydrogen and fluorine:



Test Yourself:

1. Draw the Lewis diagram for each compound.
 - A. a molecule composed of one chlorine atom and one fluorine atom
 - B. a molecule composed of one hydrogen atom and one iodine atom

Lesson 20 Unit 4 Covalent Bonds Answers

LP Steffe



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