

Chemistry – Unit 7 Worksheet 3

Write balanced chemical equations for the following reactions.

Hydrogen gas (collected over water) is also collected in the (usual) 1.0 volume flask (collected) over water.

1. Acetylene (C_2H_2) reacts with water hydrogen chloride for three acetylene, a hydrogen.
 $\text{H}_2\text{C} + \text{H}_2\text{O} \rightarrow \text{H}_2\text{C}_2\text{H}_2$

2. Calcium carbonate is decomposed (superheating) for three calcium oxide and carbon dioxide.
 $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

3. Barium sulfate reacts with water for three barium hydroxide.
 $\text{BaSO}_4 + \text{H}_2\text{O} \rightarrow \text{Ba(OH)}_2$

4. Acetylene (C_2H_2) reacts with water for three hydrogen (C_2H_2) and carbon monoxide.
 $\text{C}_2\text{H}_2 + \text{H}_2\text{O} \rightarrow \text{C}_2\text{H}_4 + \text{CO}$

5. Zinc reacts with sulfur (10) collected for three zinc sulfide (10) collected.
 $\text{Zn} + \text{S} \rightarrow \text{ZnS}$

6. Calcium sulfate decomposes sulfur for three calcium oxide and sulfur dioxide.
 $\text{CaSO}_4 \rightarrow \text{CaO} + \text{SO}_2$

7. Zinc reacts with sulfur (10) collected for three zinc sulfide and hydrogen gas.
 $\text{Zn} + \text{H}_2\text{S} \rightarrow \text{ZnS} + \text{H}_2$

8. A hydrogen containing carbon compound, $\text{C}_2\text{H}_4\text{H}_2$, decomposes for three carbon, C_2H_4 , and hydrogen gas.
 $\text{C}_2\text{H}_4\text{H}_2 \rightarrow \text{C}_2\text{H}_4 + \text{H}_2$

9. Phosphorus, P_4O_{10} , is formed when carbon monoxide reacts with phosphorus gas.
 $\text{P}_4 + \text{CO} \rightarrow \text{P}_4\text{O}_{10}$

Modeling Chemistry Unit 7 Ws3 Answers

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Table of Contents Modeling Chemistry Unit 7 Ws3 Answers

1. Understanding the eBook Modeling Chemistry Unit 7 Ws3 Answers
 - The Rise of Digital Reading Modeling Chemistry Unit 7 Ws3 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry Unit 7 Ws3 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Chemistry Unit 7 Ws3 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry Unit 7 Ws3 Answers
 - Personalized Recommendations
 - Modeling Chemistry Unit 7 Ws3 Answers User Reviews and Ratings
 - Modeling Chemistry Unit 7 Ws3 Answers and Bestseller Lists
5. Accessing Modeling Chemistry Unit 7 Ws3 Answers Free and Paid eBooks
 - Modeling Chemistry Unit 7 Ws3 Answers Public Domain eBooks
 - Modeling Chemistry Unit 7 Ws3 Answers eBook Subscription Services
 - Modeling Chemistry Unit 7 Ws3 Answers Budget-Friendly Options

6. Navigating Modeling Chemistry Unit 7 Ws3 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Modeling Chemistry Unit 7 Ws3 Answers Compatibility with Devices
 - Modeling Chemistry Unit 7 Ws3 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modeling Chemistry Unit 7 Ws3 Answers
 - Highlighting and Note-Taking Modeling Chemistry Unit 7 Ws3 Answers
 - Interactive Elements Modeling Chemistry Unit 7 Ws3 Answers
8. Staying Engaged with Modeling Chemistry Unit 7 Ws3 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry Unit 7 Ws3 Answers
9. Balancing eBooks and Physical Books Modeling Chemistry Unit 7 Ws3 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry Unit 7 Ws3 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modeling Chemistry Unit 7 Ws3 Answers
 - Setting Reading Goals Modeling Chemistry Unit 7 Ws3 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modeling Chemistry Unit 7 Ws3 Answers
 - Fact-Checking eBook Content of Modeling Chemistry Unit 7 Ws3 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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