



Remote control 10P Connector

Pin No.	Name	Pin No.	Name
1	REMO-IN	6	REMO-OUT
2	—	7	Telephone Mute
3	Telephone Signal	8	Telephone GND
4	M-SCK	9	M-BUSY
5	M-DATA	10	Shield Earth(M-BUS)

Power supply 14P Connector

Pin No.	Name	Pin No.	Name
1	Speaker FR (+)	8	Speaker FL (-)
2	Speaker FL (+)	9	Illumination (-)
3	Illumination (+)	10	Battery (+)
4	Antenna + B	11	Accessory (+)
5	Speaker RL (+)	12	—
6	Speaker RR (+)	13	Speaker RL (-)
7	Speaker FR (-)	14	Speaker RR (-)

CD Changer 13P DIN Socket

Pin No.	Name	Pin No.	Name
1	Signal GND	8	—
2	Lch	9	Battery (+)
3	—	10	Illumination Level
4	Rch	11	M-BUSY
5	—	12	M-SCK
6	—	13	M-DATA
7	ACC (+)	H	Shield Earth

Mitsubishi Galant Stereo Code

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Table of Contents Mitsubishi Galant Stereo Code

1. Understanding the eBook Mitsubishi Galant Stereo Code
 - The Rise of Digital Reading Mitsubishi Galant Stereo Code
 - Advantages of eBooks Over Traditional Books
2. Identifying Mitsubishi Galant Stereo Code
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mitsubishi Galant Stereo Code
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mitsubishi Galant Stereo Code
 - Personalized Recommendations

- Mitsubishi Galant Stereo Code User Reviews and Ratings
- Mitsubishi Galant Stereo Code and Bestseller Lists
- 5. Accessing Mitsubishi Galant Stereo Code Free and Paid eBooks
 - Mitsubishi Galant Stereo Code Public Domain eBooks
 - Mitsubishi Galant Stereo Code eBook Subscription Services
 - Mitsubishi Galant Stereo Code Budget-Friendly Options
- 6. Navigating Mitsubishi Galant Stereo Code eBook Formats
 - ePub, PDF, MOBI, and More
 - Mitsubishi Galant Stereo Code Compatibility with Devices
 - Mitsubishi Galant Stereo Code Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mitsubishi Galant Stereo Code
 - Highlighting and Note-Taking Mitsubishi Galant Stereo Code
 - Interactive Elements Mitsubishi Galant Stereo Code
- 8. Staying Engaged with Mitsubishi Galant Stereo Code
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mitsubishi Galant Stereo Code
- 9. Balancing eBooks and Physical Books Mitsubishi Galant Stereo Code
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mitsubishi Galant Stereo Code
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mitsubishi Galant Stereo Code
 - Setting Reading Goals Mitsubishi Galant Stereo Code
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mitsubishi Galant Stereo Code
 - Fact-Checking eBook Content of Mitsubishi Galant Stereo Code

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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1 types of obd2 protocols 1 1 sae j1850 pwm 1 2 sae j1850 vpw 1 3 iso 9141 2 1 4 iso 14230 kwp2000 1 5 iso 15765 4 can sae j2480 2 what are the ways to check my obd2 protocol 3 which obd2 protocol is supported by my vehicle 4 bottom line types of obd2 protocols below you will find obd2 protocols by vehicle make and model sae j1850 pwm

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obd ii pids on board diagnostics parameter ids are codes used to request data from a vehicle used as a diagnostic tool sae standard j1979 defines many obd ii pids all on road vehicles and trucks sold in north america are required to support a subset of these codes primarily for state mandated emissions inspections

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dec 14 2015 protocol used on pre 2005 nissan vehicles supported by rencom some vehicles may require manual wire connection to obd2 or nissan consult 14pin connector obd2 vehicles with ddl ecus have connected pins 12 and 13in obd ii nissan terrano 2000 obd ii no special requirements abs airbag

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obd ii changes things by making them the same what is the obd ii standard obd ii standardizes communication for the data link connector shape and pinout trouble codes and communication protocols obd ii allows any generic scan tool to connect and talk to any car this is good news for non specializing technicians

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proprietary can protocols only accessible through direct connection to can bold obdii protocols tested by aim italic suggested obdii protocols manufacturer model year obdii connection quick connection to obdii plug communication through proprietary can protocols if available or obdii standard protocols canbus direct connection

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jan 26 2004 the european union obd legislation is somewhat more complicated an obd ii compliant vehicle can use any of the five communication protocols j1850 pwm j1850 vpw iso9141 2 iso14230 4 also known as keyword protocol 2000 and more recently iso15765 4 sae j2480 a flavor of can

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abs consult ii protocol abs module srs consult ii protocol airbag module ecmd consult ii protocol diesel engine control module obdii obdii protocol engine control module wbo2 innovate motorsports wide band oxygen sensor due to limitations of consult ii protocol only one ecu can be connected at the time it s not

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obd ii protocols there are five different communication protocols available under the obd ii spec like so many things manufacturers tend to have their preferences and think their protocol is best hence the variation here s a quick overview of each and a description of the pins used on the dlc for each sae j1850 pwm

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the

reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no₂⁺ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of nitric acid and 4.0 ml of sulfuric acid and will then react with 4.5 ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1

h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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