#### **SAMPLE PAPER (2023 -24)**

#### CHEMISTRY THEORY (043)

Max. Marks:70

Time: 3 hours

General Instructions:

Read the following instructions carefully.

- (a) There are 33 questions in this question paper with internal choice.
- (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
- (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
- (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
- (e) SECTION D consists of 2 case based questions carrying 4 marks each.
- (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
- (q) All questions are compulsory.
- (h) Use of log tables and calculators is not allowed.

#### SECTION A

The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.

- 1. Which of the following solutions will have the highest conductivity at 298 K?
  - (a) 0.01 M HCl solution

(b) 0.1 M HCl solution

(c) 0.01 M CH 3COOH solution

(d) 0.1 M CH 3COOH solution

Identify A and B:

- (a) A = 1-phenylethanal, B = acetophenone
- (b) A = Benzophenone B = formaldehyde
- (c) A= Benzaldehyde, B = Acetophenone
- (d) A = Benzophenone, B = Acetophenone
- The vitamins which can be stored in our body are:
  - (a) Vitamin A, B, D and E

(d) Vitamin A, C, D and K

(c) Vitamin A, B, C and D

- (d) Vitamin A, D, E and K
- 4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give

CH3 CH= C(CH3)COONa and yellow precipitate of CHI3?

(a) 3-Methylpent-3-en-2one

(b) 3-Methylbut-2-en- one

(c) 2, 3-Dimethylethanone

(d) 3-Methylpent-4-one

- 5. Which of the following is not correct?
  - (a) In haloarenes, the electron pairs on halogen atom are in conjugation with  $\pi$ -electrons of the ring.
  - (b) The carbon-magnesium bond is covalent and non-polar in nature .
  - (c) During S<sub>N</sub><sup>1</sup> reaction, the carbocation formed in the slow step being sp<sup>2</sup> hybridised is planar.
  - (d) Out of CH2= CH-Cl and C6H3CH2Cl, C6H3CH2Cl is more reactive towards SN1 reaction

# **June 2chemistry 970paper 2question**

**JG Myers** 

June 2chemistry 970paper 2question:

The book delves into June 2chemistry 970paper 2question. June 2chemistry 970paper 2question is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and indepth insights into June 2chemistry 970paper 2question, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
  - Chapter 1: Introduction to June 2chemistry 970paper 2question
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  - Chapter 3: June 2chemistry 970paper 2question in Everyday Life
  - ∘ Chapter 4: June 2chemistry 970paper 2question in Specific Contexts
  - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of June 2chemistry 970paper 2question. The second chapter will elucidate the essential principles that need to be understood to grasp June 2chemistry 970paper 2question in its entirety.
- 4. In chapter 3, the author will examine the practical applications of June 2chemistry 970paper 2question in daily life. The third chapter will showcase real-world examples of how June 2chemistry 970paper 2question can be effectively utilized in everyday scenarios.
- 5. In chapter 4, the author will scrutinize the relevance of June 2chemistry 970paper 2question in specific contexts. The fourth chapter will explore how June 2chemistry 970paper 2question is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about June 2chemistry 970paper 2question. The final chapter will summarize the key points that have been discussed throughout the book.
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