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Lab - Evidence for Chemical Change

One way of knowing that a chemical change has occurred is to observe that the properties of the products are different from those of the needants. The new product can their became a reaction; in another chemical reaction. In this expeniment you will observe a sequence of changes that occur when a solution that begins as copper (III) nitrate is treated with a series of different reactions. All of the reactions will take place in the same test tube. At each step you will look for evidence that a new substance is formed as a result of a chemical change. You will also observe heat changes and relate them to chemical reactions.

Provided:

- List four types of observations (evidence) that indicate when a chemical change has socured.
- For each reaction: IDENTIFY the type of chemical reaction, PRESECT the products, IDENTIFY the states of each substance, BALANCE the equation.

___ Cu()V()() () + ___ Ne()H() -- |

Cu()()H()() + | H()() -- |

Cu()() + ___ H()() -- |

Cu()() + ___ H()() -- |

Cu()() + ___ A(()) -- |

Resection Type:

____ HO(0+ ____ A(1)+

Reaction Type:

Lab Evidence For Chemical Change Answers

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