

Lewis Structure Worksheet

1. Draw the Lewis Structure for each of the following elements.

Element	Lewis Structure	Element	Lewis Structure	Element	Lewis Structure
Lithium	Li	Nitrogen	N	Silicon	Si
Beryllium	Be	Fluorine	F	Neon	Ne

2. Draw the Lewis Dot Diagram for each of the following elements. Determine how many electrons the atom has to gain or lose to become stable (fill or lose outer orbit to make 8). Determine the charge that the ion will have (recall: gain electrons = "-" charge, lose electrons = "+" charge).

Element	Lewis Structure	Needed for Stability	Charge on ion
Sodium	Na	Lose 1	+1
Oxygen	O	Gain 2	-2
Aluminum	Al	Lose 3	+3
Helium	He	N/A	N/A
Potassium	K	Lose 1	+1
Carbon	C	4	Does not form an ion

Lewis Dot Diagrams Practice Worksheet

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Lewis Dot Diagrams Practice Worksheet:

Introduction to Chemistry, Laboratory Manual T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

15 PGT Chemistry Test Papers EMRS Mocktime Publication, EMRS Exam Teachers PGT Chemistry Test Papers 15 Practice Papers Tier 1 Eklavya Model Residential Schools as per Official Exam Pattern and Syllabus **General Chemistry Workbook** Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Assessment in Science** Daniel P. Shepardson, 2001-07-31 A resource for administrators and staff developers interested in designing professional development programs and for science teachers looking for techniques and examples of classroom based assessments

Journal of Veterinary Allergy and Clinical Immunology, 1993 **Index to Media and Materials for the Mentally Retarded, Specific Learning Disabled, Emotionally Disturbed** National Information Center for Special Education Materials, 1978 **Government Reports Announcements & Index**, 1989

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