

- Directions:**  
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.  
 2. An atom with a positive charge is called a cation.  
 3. An atom with a negative charge is called an anion.  
 4. The number of protons in an atom is the atomic number.

# Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.  
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained	Charge
1	Sodium	$\text{Na}^+$	11	10	lost one	positive
2	Chlorine	$\text{Cl}^-$	17	18	gained one	negative
3	Calcium	$\text{Ca}^{2+}$	20	18	lost two	positive
4	Oxygen	$\text{O}^{2-}$	8	10	gained two	negative
5	Aluminum	$\text{Al}^{3+}$	13	10	lost three	positive
6	Sulfur	$\text{S}^{2-}$	16	18	gained two	negative
7	Hydrogen	$\text{H}^+$	1	0	lost one	positive
8	Lithium	$\text{Li}^+$	3	2	lost one	positive
9	Fluorine	$\text{F}^-$	9	10	gained one	negative
10	Magnesium	$\text{Mg}^{2+}$	12	10	lost two	positive
11	Strontium	$\text{Sr}^{2+}$	38	36	lost two	positive
12	Barium	$\text{Ba}^{2+}$	56	54	lost two	positive
13	Strontium	$\text{Sr}^{2+}$	38	36	lost two	positive
14	Calcium	$\text{Ca}^{2+}$	20	18	lost two	positive
15	Sodium	$\text{Na}^+$	11	10	lost one	positive
16	Chlorine	$\text{Cl}^-$	17	18	gained one	negative
17	Fluorine	$\text{F}^-$	9	10	gained one	negative
18	Hydrogen	$\text{H}^+$	1	0	lost one	positive
19	Lithium	$\text{Li}^+$	3	2	lost one	positive
20	Sodium	$\text{Na}^+$	11	10	lost one	positive
21	Calcium	$\text{Ca}^{2+}$	20	18	lost two	positive

# **Ion And Isotope Practice Set Answer Key**

**Lingjun Ying**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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## **Table of Contents Ion And Isotope Practice Set Answer Key**

1. Understanding the eBook Ion And Isotope Practice Set Answer Key
  - The Rise of Digital Reading Ion And Isotope Practice Set Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Ion And Isotope Practice Set Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ion And Isotope Practice Set Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ion And Isotope Practice Set Answer Key
  - Personalized Recommendations
  - Ion And Isotope Practice Set Answer Key User Reviews and Ratings
  - Ion And Isotope Practice Set Answer Key and Bestseller Lists
5. Accessing Ion And Isotope Practice Set Answer Key Free and Paid eBooks

- Ion And Isotope Practice Set Answer Key Public Domain eBooks
- Ion And Isotope Practice Set Answer Key eBook Subscription Services
- Ion And Isotope Practice Set Answer Key Budget-Friendly Options
- 6. Navigating Ion And Isotope Practice Set Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Ion And Isotope Practice Set Answer Key Compatibility with Devices
  - Ion And Isotope Practice Set Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ion And Isotope Practice Set Answer Key
  - Highlighting and Note-Taking Ion And Isotope Practice Set Answer Key
  - Interactive Elements Ion And Isotope Practice Set Answer Key
- 8. Staying Engaged with Ion And Isotope Practice Set Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ion And Isotope Practice Set Answer Key
- 9. Balancing eBooks and Physical Books Ion And Isotope Practice Set Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ion And Isotope Practice Set Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Ion And Isotope Practice Set Answer Key
  - Setting Reading Goals Ion And Isotope Practice Set Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ion And Isotope Practice Set Answer Key
  - Fact-Checking eBook Content of Ion And Isotope Practice Set Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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