Ch 13 Study Guide (Chemical Bonds)

Fill in the table below with the correct values:

| Element Name | lon Symbol | # Protons | # Electrons | Charge | Ion Type |
|-----------------|------------------|-----------|----------------|--------|----------|
| Fluorine | F. | 9 | 10 | -1 | Anion |
| lodine | I | 53 | 54 | -1 | Anien |
| Sulfur | 52- | 16 | 18 | -2 | Anion |
| Potassium | K+ | 19 | 18 | +1 | Cation |
| Calcium | Ca ²⁺ | 20 | 18 | +2 | Carion |
| Bromine | Be | 35 | 36 | -1 | Anion |
| Stroutium | Sr2+ | 38 | 36 | +2 | Cation |
| Oxygen | 03- | 8 | 10 | -2 | Anien |
| Magnesium | Mast | 12 | 10 | +2 | Cation |
| Aluminum | Al 31 | 13 | 10 | +3 | Carion |
| Selenium | Sea- | 34 | 36 | -2 | Anion |
| Lithium | Li* | 3 | 2 | +1 | Carrion |
| Rubidium | Rb+ | 37 | 36 | +1 | Cation |
| Chlorine | ce- | 17 | 18 | -1 | Anion |

2. An ionic bond is formed between what two types of atoms?

Metal & non metal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.

Mg . _ => Mg :0:

5. What is the formula for the compound formed in question #4?

Ions Answer Key

Gabi N Waite, Maria Sheakley

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