

Ionic & Covalent Compound Naming**Race**

First, identify whether these compounds are ionic or covalent. Then, use the correct formula writing rules to write the correct chemical formulas for each compound.

1)	Compound Name	Type of Compound: Ionic or Covalent	Chemical Formula
	copper (II) chlorite		Cu(ClO)
2)	sodium hydroxide		NaOH
3)	nitrogen dioxide		NO ₂
4)	cobalt (III) oxalate		Co ₂ (C ₂ O ₄) ₃
5)	ammonium sulfide		(NH ₄) ₂ S
6)	aluminum cyanide		Al(CN) ₃
7)	carbon disulfide		CS ₂
8)	tetraphosphorous pentoxide		P ₄ O ₅
9)	potassium permanganate		KMnO ₄
10)	manganese (III) chloride		MnCl ₃
11)	Compound Name	Type of Compound: Ionic or Covalent	Chemical Formula
	calcium bromate		Ca(BrO ₃) ₂
12)	carbon monoxide		CO
13)	potassium oxide		K ₂ O
14)	antimony tribromide		SbBr ₃
15)	zinc phosphate		Zn ₃ (PO ₄) ₂
16)	copper (II) bicarbonate		Cu(HCO ₃) ₂
17)	dinitrogen tetroxide		N ₂ O ₄

Ionic And Covalent Nomenclature Ws Answers

Joan I. Senyk, Darrell D. Ebbing



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