

Name Solutions

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Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers $1e^-$ to F	Li^+
F	7 ve	accepts $1e^-$ from Li	F^-

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $Li + Cl \Rightarrow LiCl$	Li	1 ve	donates $1e^-$	Li^+
$Li + Cl \rightarrow [Li]^+ [Cl:]^{-1}$	Cl	1 ve	accepts $1e^-$	Cl^-
2) $Ca + O \Rightarrow CaO$	Ca	2 ve	donates $2e^-$	Ca^{+2}
$Ca + O \rightarrow [Ca]^{+2} [O:]^{-2}$	O	6 ve	accepts $2e^-$	O^{+2}
3) $Be + F \Rightarrow BeF_2$	Be	2 ve	donates $2e^-$ to each F	Be^{+2}
$Be + F \rightarrow [Be]^{+2} [F:]^{-1}$	F	7 ve	each F accepts $1e^-$ each	F^-
4) $Mg + S \Rightarrow MgS$	Mg	2 ve	donates $2e^-$ to S	Mg^{+2}
$Mg + S \rightarrow [Mg]^{+2} [S:]^{-2}$	S	6 ve	accepts $2e^-$	S^{+2}
5) $K + F \Rightarrow KF$	K	1 ve	donates $1e^-$	K^+
$K + F \rightarrow [K]^+ [F:]^{-1}$	F	7 ve	accepts $1e^-$	F^-

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Understanding Solids Richard J. D. Tilley, 2005-09-27 A modern introduction to the subject taking a unique integrated approach designed to appeal to both science and engineering students Covering a broad spectrum of topics this book includes numerous up to date examples of real materials with relevant applications and a modern treatment of key concepts The science bias allows this book to be equally accessible to engineers chemists and physicists Carefully structured into self contained bite sized chapters to enhance student understanding Questions have been designed to reinforce the concepts presented Includes coverage of radioactivity Reflects a rapidly growing field from the science perspective **Inorganic Reactions and Methods, The Formation of Bonds to O, S, Se, Te, Po (Part 2)** A. D. Norman, 2009-09-17 Die langerwartete Komplettierung der Reihe Inorganic Reactions and Methods beginnt jetzt mit der Publikation des 6 Bandes Seit Jahren vielfach in Forschung und Berufsalltag bew hrt behandelt die Reihe alle Gebiete der anorganischen Synthesechemie die Chemie der Elemente Koordinationsverbindungen Donor Acceptor Addukte organometallische Stoffe Polymere und Festk rper sowie bioanorganische Materialien Mit Autoren Sach und Verbindungsregister 07 98 **Metals and Materials** ,1973 Focus on Physical Science California Edition Michael J. Padilla, 2007 *Intermetallic Compounds: Principles* Jack Hall Westbrook, Robert Louis Fleischer, 1995 *Encyclopedia of Inorganic Chemistry: T-Z* R. Bruce King, 2005 *Nuclear Science Abstracts* ,1960 **Chemistry** Patrick Kavanah, 2004-08-30 Solutions to Black Exercises Roxy Wilson, 2005-05 Features detailed step by step solutions to the more than 1100 black numbered end of character problems in Chemistry the central science Comprehensive Dissertation Index, 1861-1972: Chemistry Xerox University Microfilms, 1973 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Transactions of the American Institute of Mining, Metallurgical and Petroleum Engineers** American Institute of Mining, Metallurgical, and Petroleum Engineers, 1964 *Journal of the Chemical Society* ,1958 **Metallurgical Abstracts** ,1960 *The NMR Relaxation Mechanisms of O17 in Aqueous Solutions of Paramagnetic Cations and the Lifetime of Water Molecules in the First Coordination Sphere* Terrence James Swift, 1962 *Physical Chemistry* R. Stephen Berry, Stuart Alan Rice, John Ross, 1980 **Handbook on the Physics and Chemistry of Rare Earths** Karl A. Gschneidner (Jr.), LeRoy Eyring, 1978 The rare earths play a unique role in science These seventeen related elements afford a panoply of subtle variations deriving from the systematic development of their electronic configurations allowing a test of theory with excellent resolution In contrast they find widespread use in even the most mundane processes such as steel making for polishing materials and gasoline cracking

catalysts In between are exotic uses such as TV screen phosphors lasers high strength permanent magnets and chemical probes This multi volume handbook covers the entire rare earth field in an integrated manner Each chapter is a comprehensive up to date critical review of a particular segment of the field The work offers the researcher and graduate student alike a complete and thorough coverage of this fascinating field **Journal of the Chemical Society** Chemical Society (Great Britain),1958-11 **Quarterly Journal of the Chemical Society of London** Chemical Society (Great Britain),1971 Physical Chemistry: Matter in equilibrium R. Stephen Berry,Stuart Alan Rice,John Ross,1980

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