10 Honors Ionic and Metallic Bonding Questions [21 marks]

1. [1 mark]

Which is correct for all solid ionic compounds?

A. High volatility

B. Poor electrical conductivity

C. Low melting point

D. Good solubility in water

2. [1 mark]

A. B. C.

Which combination corresponds to a strong metallic bond? B

Charge on the metal ion	Radius of ion
large	large
large	small
small	small
small	large

3. [I mark]

How does a lithium atom form the most stable ion?

A. The atom gains a proton to form a positive ion.

B. The atom loses a proton to form a negative ion.

C. The atom loses an electron to form a positive ion.

D. The atom gains an electron to form a negative ion.

4. [I mark]

What is the formula of ammonium phosphate?

Ionic And Metallic Bonding Questions

William D. Callister, Jr., David G. Rethwisch

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