

Chemistry: Ionic & Covalent Bonds Unit Study Guide

Covalent Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Covalent Compounds.

- Bond (ion)
- (Large / Small) Electronegativity Difference
- (Same / Different) Elements
- (High / Low) Melting & Boiling Point
- (Conducts / Does not Conduct) Electricity in Solution
- (Solid / Liquid / Gas) at Room Temperature

2. Write the prefix following number

Atoms	Prefix
1	
2	
3	
4	
5	

4. What are the two

5. Complete the chart

Formula	Name
BF_3	
	Fe
S_8	
	C

6. What does VSEPR

Name: _____ Date: _____ Period: _____

Ionic Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Ionic Compounds.

- Bond between a (Metal / Nonmetal) and (Metal / Nonmetal)
- (Large / Small) Electronegativity Difference
- The (Metals / Nonmetals) form Cations, which (Gain / Lose) Electrons
- The (Metals / Nonmetals) form Anions, which (Gain / Lose) Electrons
- (High / Low) Melting & Boiling Point
- (Conducts / Does not Conduct) Electricity in Solution
- (Solid / Liquid / Gas) at Room Temperature

2. What charge would the following periodic table families have in a chemical bond?

Family	Charge
Alkali Metals	
Alkali Earths	
Transition	
Carbon	
Nitrogen	
Halogens	

3. Write the roman numeral that is represented by the following numbers.

Roman Numeral	Number
	1
	2
	3
	4
	5
	6

4. Complete the chart below by filling in the missing boxes.

CATION	ANION	FORMULA	NAME
Mg^{2+}	Br^{-}		
		CuO	
			Iron (III) chloride
		$(NH_4)_3PO_4$	

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