



# CHEMICAL BONDS



## IONIC, COVALENT, & METALLIC

**FORMS OF CHEMICAL BONDS**  
Bonding: combination of atoms

Atoms combine to form molecules or compounds by sharing or transferring electrons. This process is called bonding. The resulting molecule or compound has properties different from the individual atoms.

**Types of Bonds:**

- Covalent Bond:** Formed by sharing electrons between two atoms. Example:  $H_2$ ,  $H_2O$ ,  $CO_2$ .
- Ionic Bond:** Formed by the transfer of electrons from one atom to another, creating oppositely charged ions that attract each other. Example:  $NaCl$ ,  $MgO$ .
- Metallic Bond:** Formed by the attraction between positive metal ions and a sea of delocalized electrons. Example:  $Fe$ ,  $Cu$ ,  $Al$ .

**Covalent, Ionic, and Metallic Bonds**

Covalent bonds are formed by sharing electrons. They are typically found in nonmetals and their compounds. Ionic bonds are formed by the transfer of electrons. They are typically found in metals and nonmetals. Metallic bonds are formed by the attraction between positive metal ions and a sea of delocalized electrons. They are typically found in metals.

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# Ionic And Metallic Bonding Guided Reading

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**Study Guide for Atoms, Molecules, and Life** Michael S. Matta, 1981 Readers' Guide to Periodical Literature, 1941 **Prentice Hall Chemistry**, 2000 *Chemical Interactions*, 2005 **GGSIU B.Sc Hons Nursing Guide 2022** Deepa Chaudhary, Virendra, Priyanka Sharm, Rashmi Sharma, Ashwini Kumar Prasad, 2022-02-15

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**Study Guide/Selected Solutions Manual** Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **Study Guide**

**for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples      *Student's Guide to Masterton and Slowinski's Chemical Principles* Ray Boyington, William L. Masterton, 1977      **Chemistry of Matter** , 1997

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