

# IONIC BONDING

Name \_\_\_\_\_

Ionic bonding occurs when a metal transfers one or more electrons to a nonmetal in an effort to attain a stable octet of electrons. For example, the transfer of an electron from sodium to chlorine can be depicted by a Lewis dot diagram.



Calcium would need two chlorine atoms to get rid of its two valence electrons.



Show the transfer of electrons in the following combinations.



# Ionic Bonding Instructional Fair Answer Key

**University of California, Berkeley**

A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the University of California, Berkeley text.

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