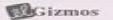
Activity A: Small atoms		Get the Gizmo ready:		
		 On the PERIODIC TABLE tab, select H (hydrogen). Select the ELECTRON CONFIGURATION tab. Click Reset. 	1 H 2	
Off	ganization are sho	rons are arranged in orbitals , subshells , and shells . These leaven by the boxes of the Gizmo. Each box represents an orbital d with letters (s , p , d , and t) and the shells are labeled with nur	. The	
Q	uestion: How are	electrons arranged in elements with atomic numbers 1 th	rough 10?	
1.	Infer: Based on	its atomic number, how many electrons does a hydrogen atom	have?	
2.	Arrange: The Aufbau principle states that electrons occupy the lowest-energy orbital. Click once in the 1s box to add an electron to the only orbital in the <i>s</i> subshell of the first shell.			
	Click Check. What is the electron configuration of hydrogen?			
3.	Arrange: Click Next element to select helium. Add another electron to the 1s orbital. The arrows represent the spin of the electron. What do you notice about the arrows?			
	The Pauli exclusion principle states that electrons sharing an orbital have opposite spins.			
4.	Check your work	Check your work: Click Check. What is the electron configuration of helium?		
5.	Arrange: Click Next element and create electron configurations for lithium, beryllium, and boron. Click Check to check your work, and then list each configuration below:			
	Lithium:	Beryllium: Boron:		
6.	Arrange: Click Next element to select carbon. Add a second electron to the first 2p orbital.			
	Click Check. What feedback is given?			
7.	Rearrange: Hund's rule states that electrons will occupy an empty orbital when it is available in that subshell. Rearrange the electrons within the 2p subshell and click Check.			
	Is the configurat	ion correct now? 1s		
	Show the correc	t configuration in the boxes at right: 2s 2p		

(Activity A continued on next page)



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Academy of Engineering, National
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The Art and Science of Questions Ronald Legarski, 2024-09-28 In The Art and Science of Questions Ronald Legarski explores the profound power of questioning in shaping human thought innovation and discovery Whether in business science education or everyday life the ability to ask the right question can lead to deeper understanding better decisions and groundbreaking ideas This book offers a comprehensive guide to mastering the art of inquiry covering various types of questions from open ended to leading reflective and hypothetical and providing practical methods for applying them in real world contexts With insights drawn from philosophy psychology leadership and technology this book equips readers with the tools to ask more effective questions and unlock the full potential of inquiry Legarski also delves into the future of questioning examining how artificial intelligence virtual reality and quantum computing will revolutionize how we ask and answer guestions in the coming decades Whether you re a student a professional or simply a curious mind The Art and Science of Questions will transform the way you approach problem solving learning and communication Key Features Detailed exploration of question types including funneling Socratic and reflective questioning Real world case studies demonstrating the impact of effective questioning across various fields Practical strategies for improving critical thinking decision making and leadership through inquiry Insight into the future of questioning with AI and emerging technologies Published by SolveForce 2024 Investigating Chemistry Matthew Johll, 2006-03-17 Matthew Johll s book introduces students from a non science background to the fundamentals of chemistry through an array of examples and applications from real life crime scenes Sherlock Holmes stories and authentic accounts of drug deals murders and thefts **Introduction to Physical Science** James T. Shipman, Jerry D. Wilson, Aaron W. Todd, 2002-07-08 This text presents a survey of the physical sciences physics chemistry astronomy meteorology and geology for non science majors Topics are treated both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any spectrum in between **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science

Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum Exploring Chemistry (Loose-Leaf) Matthew specialists advocates of hands on science teaching and concerned parents Johll, 2012-05-04 Matthew Johll's Exploring Chemistry overs the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It s the same same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book s specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more The Chemistry Steven S. Zumdahl, Susan A. Zumdahl, 2000 Software Encyclopedia 2000 Bowker Editorial Staff, 2000-05 Zumdahl s conceptual approach to problem solving early and thorough treatment of reactions and integration of descriptive chemistry and chemical principles throughout are key elements to the success of this bestseller By encouraging students to understand qualitative concepts before considering quantitative problems Chemistry helps students build a solid understanding of topics and avoid common misconceptions The text also includes new Chemical Impact boxes that present everyday applications of chemistry plus over 200 redesigned illustrations with greater clarity and emphasis on molecular level illustrations Energy Research Abstracts, 1990 Semiannual with semiannual and annual indexes References to all scientific and technical literature coming from DOE its laboratories energy centers and contractors Includes all works

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