



## Equations

Name \_\_\_\_\_

Date \_\_\_\_\_

Write a balanced chemical equation to represent each of the following chemical reactions.

1. iron + sulfur  $\rightarrow$  iron(II) sulfide
2. zinc + copper(II) sulfate  $\rightarrow$  zinc sulfate + copper
3. silver nitrate + sodium bromide  $\rightarrow$  sodium nitrate + silver bromide
4. potassium chlorate (heated)  $\rightarrow$  potassium chloride + oxygen
5. water (electricity)  $\rightarrow$  hydrogen + oxygen
6. mercury(II) oxide (heated)  $\rightarrow$  mercury + oxygen
7. potassium iodide + lead(II) nitrate  $\rightarrow$  lead(II) iodide + potassium nitrate
8. aluminum + oxygen  $\rightarrow$  aluminum oxide
9. magnesium chloride + ammonium nitrate  $\rightarrow$  magnesium nitrate + ammonium chloride
10. iron(III) chloride + ammonium hydroxide  $\rightarrow$  iron(III) hydroxide + ammonium chloride
11. sodium peroxide + water  $\rightarrow$  sodium hydroxide + oxygen
12. iron(III) oxide + carbon  $\rightarrow$  iron + carbon monoxide
13. iron + water  $\rightarrow$  hydrogen + iron(III) oxide
14. iron(III) chloride + potassium hydroxide  $\rightarrow$  potassium chloride + iron(III) hydroxide
15. aluminum + sulfuric acid  $\rightarrow$  aluminum sulfate + hydrogen
16. sodium carbonate + calcium hydroxide  $\rightarrow$  sodium hydroxide + calcium carbonate



# J Weston Walch Publisher 13 Bases Crossword Chemistry

**Brian Pressley**



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