

Managerial Accounting Midterm 1 Review

Chapter 1: Managerial Accounting

Managerial Accounting Basics:

Managerial Accounting: a field of accounting that provides economic and financial information for managers and other stakeholders

Differences Between Financial and Managerial Accounting

Features	Financial Accounting	Managerial Accounting
Primary users & objectives	External users: shareholders, creditors, and lenders	Internal users: different parts of management
Types and frequency of reports	Standardized reports	Customized as needed
Purpose of reports	General purpose	Specific: for particular decisions
Content of reports	Historical financials as an indicator of firm's performance	Historical financials and other accounting data for decision making
Regulation & standards	Standard by IASB (International Accounting Standards)	No independent bodies

Management Functions and Organizational Structures:

Group of functions: the group of activities shared by the three functions of management at each grade of organization to formulate operating policies, select activities, coordinate them through the company

Staff functions: officers (SFOs) who provide advice but do not have direct responsibility for managing the business; but are also designated with direct responsibility; officers who provide advice

Line functions: jobs that are directly involved in a company's main business providing sustaining activities

Staff functions: jobs that support the efforts of line employees

Chief Financial Officer (CFO): the corporate officer who is responsible for all company's accounting and finance issues

Controller: the financial officer who is responsible for a company's accounting records, system of internal controls, and preparation of financial statements, tax returns, and internal reports

Managerial Accounting Midterm

Ronald M. Copeland, Paul E. Dascher



Managerial Accounting Midterm:

Cost/managerial Accounting Exam Questions and Explanations Irvin N. Gleim, 2009 *Cost/managerial Accounting* Irvin N. Gleim, Terry L. Campbell, 1998 **CIMA Official Learning System Test of Professional Competence in Management Accounting** Heather Barnwell, Nick Best, 2008-07-15 The 2009 edition of CIMA's Official Learning Systems has been written in conjunction with the Examiner to fully reflect what could be tested in the exam Fully revised and in 2 color paperback format the 2009 Learning Systems provide complete study material for the May and November 2009 exams This edition includes practice questions throughout complete revision section topic summaries recommended reading articles from a range of journals Q A's CIMA Learning Systems are the only study materials endorsed and recommended by CIMA

Managerial Accounting Jerry J. Weygandt, Paul D. Kimmel, Donald E. Kieso, 2009-10-19 Many corporate managers struggle to see the relevance of accounting in their everyday responsibilities Weygandt shows them how managerial accounting information fits in the larger context of business so they are better able to understand the important concepts The new Do It feature reinforces the basics by providing quick hitting examples of brief exercises The chapters also incorporate the All About You AAY feature as well as the Accounting Across the Organization AAO boxes that highlight the impact of accounting concepts With these features readers will have numerous opportunities to think about what they have just read and then apply that knowledge to sample problems *Management Accounting Fundamentals May 2003 Exam Questions and Answers* J. Walker, 2003-06 Examiners lecturers and students know that practice is vital to exam success so why not test your knowledge with these useful books Management Accounting theory and practice M P Pandikumar, 2009

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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example $\text{hcl} + \text{naoh} \rightarrow \text{nacl} + \text{h}_2\text{o}$

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web during this reaction the h^+ ions from the acid and the oh^- ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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web other articles where exothermic reaction is discussed alkali metal reactions with nonmetals the reactions are highly
exothermic producing up to 235 kcal mole for lithium fluoride the alkali metals react with nonmetals in groups 15 and 16 va
and via of the periodic table sulfides can be formed by the direct reaction of the alkali metals with

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web 1 sodium hydroxide solution is poured into a beaker of hydrochloric acid which contains a thermometer showing room
temperature 2 the beaker now contains sodium chloride and water and the

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable
measure an appropriate volume of each liquid eg 25 cm³ place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less
exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of
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web indicators are used to determine whether a solution is acidic or alkaline acids react with metals bases and carbonates to
produce salts neutralisation is the reaction between an acid and a base

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic
eg 1 HCl(aq) + NaOH(aq) → NaCl(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹ eg 2 HNO₃(aq) + KOH(aq) → KNO₃(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o aq}$ you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts

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