

- 2 In an experiment, powdered calcium hydroxide was added to dilute hydrochloric acid and the pH was measured.

The method used was

- step 1** measure  $200\text{ cm}^3$  dilute hydrochloric acid into a beaker  
**step 2** add 0.1 g of powdered calcium hydroxide to the beaker  
**step 3** find the pH of the mixture  
**step 4** repeat steps 2 and 3 until the pH stops changing.

- (a) State what should be done after **step 2** to make sure that any reaction is complete.

(1)

- (b) Complete the word equation for the reaction.

(2)

calcium hydroxide + hydrochloric acid  $\rightarrow$  \_\_\_\_\_

- (c) Which row of the table shows the state symbols for powdered calcium hydroxide

		calcium hydroxide	hydrochloric acid
(1)	(2)	(3)	(4)
(1)	(2)	(3)	(4)
(1)	(2)	(3)	(4)
(1)	(2)	(3)	(4)

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**L Reisser**



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