

## CHEM 1201 - Mastering Chemistry HW 1 & 2 + Exam 1 Review (Ch. 1-3, Prof. Elgrishi) study guide verified A+ tips NEW!!!

- 1. 2, 1:** What is the ratio of hydrogen atoms (HH) to oxygen atoms (OO) in 2 LL of water? Enter the simplest whole number ratio in order of hydrogen to oxygen, respectively.
- 2. 600 H atoms:** How many atoms of hydrogen (HH) are present in 200 molecules of ammonia ( $\text{NH}_3$ )?
- 3.  $8.35 \times 10^{25}$  H atoms:** A solution of ammonia and water contains  $2.90 \times 10^{25}$  water molecules and  $8.50 \times 10^{24}$  ammonia molecules. How many total hydrogen atoms are in this solution?
- 4. Intensive:**
  - matte grey in color
  - begins to melt at 961.78 degrees Celcius
  - has a density of 10.49 g/mL**Extensive:**
  - has a volume of 7.42 mL
  - has a mass of 77.83 g**Classify the following as intensive or extensive properties of silver.**
  - matte grey in color
  - begins to melt at 961.78 degrees Celcius
  - has a density of 10.49 g/mL
  - has a volume of 7.42 mL
  - has a mass of 77.83 g
- 5. the density of the sample**  
**the melting point of the sample:** You are given a sample resembling silver. What properties could be used to help determine whether the sample is really silver?
- 6. 189 g:** By definition,  
 $\text{density} = \text{mass} / \text{volume}$   
Typical units of density are g/mL or  $\text{g/cm}^3$ .  
 $1\text{mL} = 1\text{cm}^3$   
The density of iron is 7.87 g/mL. What is the mass of a  $4.00\text{ cm} \times 3.00\text{ cm} \times 2.00\text{ cm}$  block of iron?
- 7. In moist air it combines with oxygen to form aluminum oxide.**

# Mastering Chemistry Answers Ch 13

**RM Cervero**



## **Mastering Chemistry Answers Ch 13:**

## Unveiling the Magic of Words: A Overview of "**Mastering Chemistry Answers Ch 13**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Mastering Chemistry Answers Ch 13**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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