

Unit 1 Chemistry Study Guide

History of the Atom:

Democritus (450 BCE):

- Matter is not infinitely divisible.
- There are single particles that cannot be divided further, called **atomos**.

Aristotle (380 BCE):

- Matter is infinitely divisible.
- Matter is called **hyle**.

Antoine Lavoisier (1774):

- **Law of Conservation of Mass:** Mass cannot be created or destroyed.

Joseph Proust (1779):

- **Law of Definite Proportions:** All samples of a pure compound contain the same elements in the same proportion by mass (no matter quantity).

John Dalton (1803 – 1808):

- Observed work of Lavoisier and Proust
- Led Dalton to develop the first **atomic theory**.
- **Dalton's Atomic Theory:**
 - o All matter is composed of extremely small particles called atoms.
 - o All atoms of a given element are identical, having the same size, mass, and chemical properties. The atoms of one element are different from the atoms of all other elements.
 - o Atoms cannot be subdivided, created, or destroyed.
 - o Atoms of different elements can combine in simple whole number ratios to form chemical compounds.
 - o In chemical reactions, atoms are combined, separated or rearranged.
- **Law of Multiple Proportions:** Elements can combine in different proportions. However, the different compounds will have different properties but the ratio of atoms is always a whole number.

Billiard Ball Model:



atom

J.J. Thompson (1906):

- **Cathode Ray Experiment –**
 - o Part 1: Thompson ran electricity through a cathode ray tube and when he put the negatively charged end of a magnet near the tube, he saw the ray of electricity

Mastering Chemistry Study Guide

Ian Pickup



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Mastering Chemistry Study Guide**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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Table of Contents Mastering Chemistry Study Guide

1. Understanding the eBook Mastering Chemistry Study Guide
 - The Rise of Digital Reading Mastering Chemistry Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Mastering Chemistry Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mastering Chemistry Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mastering Chemistry Study Guide
 - Personalized Recommendations
 - Mastering Chemistry Study Guide User Reviews and Ratings
 - Mastering Chemistry Study Guide and Bestseller Lists

5. Accessing Mastering Chemistry Study Guide Free and Paid eBooks
 - Mastering Chemistry Study Guide Public Domain eBooks
 - Mastering Chemistry Study Guide eBook Subscription Services
 - Mastering Chemistry Study Guide Budget-Friendly Options
6. Navigating Mastering Chemistry Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Mastering Chemistry Study Guide Compatibility with Devices
 - Mastering Chemistry Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mastering Chemistry Study Guide
 - Highlighting and Note-Taking Mastering Chemistry Study Guide
 - Interactive Elements Mastering Chemistry Study Guide
8. Staying Engaged with Mastering Chemistry Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mastering Chemistry Study Guide
9. Balancing eBooks and Physical Books Mastering Chemistry Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mastering Chemistry Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mastering Chemistry Study Guide
 - Setting Reading Goals Mastering Chemistry Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mastering Chemistry Study Guide
 - Fact-Checking eBook Content of Mastering Chemistry Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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