

## Start Quiz 2 Material

### Leaving Group Ability

- Weakest base is best, most stable leaving group
- Find the weakest base because conjugate acid has lowest pKa

### How Carboxylic Acids and Derivatives React

#### Nomenclature

- Acyl chloride (acid chloride)



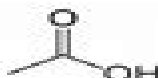
- Ester



- Acetamide



Benzoic Acid



- Carboxylic acid



- Amide



Acetic

Acid



### Relative Reactivities of Carboxylic Acids and Derivatives



- Resonance argument explains reactivities of carboxylic acids/esters versus amides because resonates to O or N with +FC. N is less EN than O so more stable with +FC. Carboxylic acids and esters are therefore more susceptible to nucleophilic addition.

### Acid Chlorides

- Reactant: acyl chloride
- Reagent: nucleophile (alcohol, water, ammonia, etc)
- Product: depends on nucleophile (ester, carboxylic acid, amide)
- Mechanism:



- So reactive that they react under neutral conditions
- No  $S_N2$  at  $sp^2$  so always write the tetrahedral intermediate
- Sequence called (1) addition-elimination or (2) nucleophilic acyl substitution

# National Standardized Organic Chemistry Test Study Guide

**JL Elias**



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