

WORKSHEET – Naming Compounds KEY

Part I – Naming Compounds and Numbering Atoms

FORMULA	NAME	# of DIFFERENT ATOMS	TOTAL # OF ATOMS
1. CaCl_2	Calcium chloride	1 Ca; 2 Cl	3
2. AgCl	<i>Silver chloride</i>	1 Ag; 1 Cl	2
3. MgO	<i>Magnesium oxide</i>	1 Mg; 1 O	2
4. NaBr	<i>Sodium bromide</i>	1 Na; 1 Br	2
5. Al_2O_3	<i>Aluminum oxide</i>	2 Al; 3 O	5
6. KI	<i>Potassium iodide</i>	1 K; 1 I	2
7. ZnCl_2	<i>Zinc chloride</i>	1 Zn; 2 Cl	3
8. Li_2O	<i>Lithium oxide</i>	2 Li; 1 O	3
9. BaO	<i>Barium oxide</i>	1 Ba; 1 O	2
10. CaBr_2	<i>Calcium bromide</i>	1 Ca; 2 Br	3
11. MgCl_2	<i>Magnesium chloride</i>	1 Mg; 2 Cl	3
12. AgI	<i>Silver iodide</i>	1 Ag; 1 I	2
13. ZnS	<i>Zinc sulphide</i>	1 Zn; 1 S	2
14. BaF_2	<i>Barium fluoride</i>	1 Ba; 2 F	3
15. Ca_3P_2	<i>Calcium phosphide</i>	3 Ca; 2 P	5
16. Na_2O	<i>Sodium oxide</i>	2 Na; 1 O	3
17. AlN	<i>Aluminum nitride</i>	1 Al; 1 N	2
18. K_3N	<i>Potassium nitride</i>	3 K; 1 N	4
19. Ca_2C	<i>Calcium carbide</i>	2 Ca; 1 C	3
20. K_2O	<i>Potassium oxide</i>	2 K; 1 O	3

Ionic Compounds Homework Review Sheet Answer Key

VM Jensen



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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs *Chemistry, Student Study*

Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **Investigating Chemistry** Matthew Johll, 2008-12-22 In its new second edition Investigating Chemistry A

Forensic Science Perspective remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media supplements package **Holt Chemistry** Ralph Thomas Myers, 2004 Science Software Quarterly, 1985

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geology 6 miscellaneous and 16 physics *Foundations of Physical Science* Tom Hsu PH.D.,2002 Chapter Resource 2
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and Theory Multiple Choice Questions 37 MCQs Benzene Chemical Compound Multiple Choice Questions 41 MCQs Carbonyl
Compounds Multiple Choice Questions 29 MCQs Carboxylic Acids and Acyl Compounds Multiple Choice Questions 27 MCQs
Chemical Bonding Multiple Choice Questions 213 MCQs Chemistry of Life Multiple Choice Questions 29 MCQs Electrode
Potential Multiple Choice Questions 62 MCQs Electrons in Atoms Multiple Choice Questions 53 MCQs Enthalpy Change
Multiple Choice Questions 45 MCQs Equilibrium Multiple Choice Questions 50 MCQs Group IV Multiple Choice Questions 53
MCQs Groups II and VII Multiple Choice Questions 180 MCQs Halogenoalkanes Multiple Choice Questions 33 MCQs
Hydrocarbons Multiple Choice Questions 53 MCQs Introduction to Organic Chemistry Multiple Choice Questions 52 MCQs
Ionic Equilibria Multiple Choice Questions 56 MCQs Lattice Energy Multiple Choice Questions 33 MCQs Moles and
Equations Multiple Choice Questions 50 MCQs Nitrogen and Sulfur Multiple Choice Questions 89 MCQs Organic and
Nitrogen Compounds Multiple Choice Questions 54 MCQs Periodicity Multiple Choice Questions 202 MCQs Polymerization
Multiple Choice Questions 36 MCQs Rates of Reaction Multiple Choice Questions 39 MCQs Reaction Kinetics Multiple Choice
Questions 52 MCQs Redox Reactions and Electrolysis Multiple Choice Questions 55 MCQs States of Matter Multiple Choice
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Table of Contents Ionic Compounds Homework Review Sheet Answer Key

1. Understanding the eBook Ionic Compounds Homework Review Sheet Answer Key
 - The Rise of Digital Reading Ionic Compounds Homework Review Sheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Compounds Homework Review Sheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Compounds Homework Review Sheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Compounds Homework Review Sheet Answer Key
 - Personalized Recommendations
 - Ionic Compounds Homework Review Sheet Answer Key User Reviews and Ratings
 - Ionic Compounds Homework Review Sheet Answer Key and Bestseller Lists
5. Accessing Ionic Compounds Homework Review Sheet Answer Key Free and Paid eBooks

- Ionic Compounds Homework Review Sheet Answer Key Public Domain eBooks
- Ionic Compounds Homework Review Sheet Answer Key eBook Subscription Services
- Ionic Compounds Homework Review Sheet Answer Key Budget-Friendly Options
- 6. Navigating Ionic Compounds Homework Review Sheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Compounds Homework Review Sheet Answer Key Compatibility with Devices
 - Ionic Compounds Homework Review Sheet Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Compounds Homework Review Sheet Answer Key
 - Highlighting and Note-Taking Ionic Compounds Homework Review Sheet Answer Key
 - Interactive Elements Ionic Compounds Homework Review Sheet Answer Key
- 8. Staying Engaged with Ionic Compounds Homework Review Sheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Compounds Homework Review Sheet Answer Key
- 9. Balancing eBooks and Physical Books Ionic Compounds Homework Review Sheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Compounds Homework Review Sheet Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ionic Compounds Homework Review Sheet Answer Key
 - Setting Reading Goals Ionic Compounds Homework Review Sheet Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ionic Compounds Homework Review Sheet Answer Key
 - Fact-Checking eBook Content of Ionic Compounds Homework Review Sheet Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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