INDORE PUBLIC SCHOOL MID TERM EXAM PRACTICE PAPER

Class-XI

SUBJECT - CHEMISTRY (THEORY)

M.M.: 70

Time: 3 Hrs

General instructions: -

- (i) This question paper contains 26 questions. Write answers neatly and legibly. All questions are compulsory. This question paper has five sections: Section A, Section B, Section C, Section D and Section E.
- (ii) Question no. 1 to 5 are very short answer questions and carry 1 mark each.
- (iii) Question no. 6 to 10 are short answer questions and carry 2 marks each.
- (iv) Question no. 11 to 22 are also short answer questions and carry 3 marks each.
- (v) Question no. 23 is value based question and carry 4 marks.
- (vi) Question no. 24 to 26 are long answer questions and carry 5 marks each.
- (vii) Use log tables, if necessary. Use of calculators is not allowed.

SECTION-A [5 Questions (1 Mark Each), Total Marks: 5]

- Q.1 How many numbers of atoms are present in 1.4 gram of nitrogen gas?
- Q.2 Write the electronic configuration of Chromium and Copper? (At. No. are 24, 29 respectively)
- Q.3 What is the n_{factor} of KMnO₄ in acidic medium?
- Q.4 Define ionization energy?
- Q.5 Give hybridisation of XeO₂F₂ and CO₃²⁻.

SECTION-B [5 Questions (2 Marks Each), Total Marks: 10]

- Q.6 Nitrogen and Hydrogen gets combined to give Ammonia 0.5 mole of N₂ and 2.0 moles of H₂ are taken initially. Find:
 - i) What is limiting reagent?
 - ii) Weight of NH3 gets formed?
- Q.7 Define Aufbau rule and Pauli's Exclusion Principle.
- Q.8 What are the reasons behind diagonal relationship and write the elements which shows the diagonal relationship?
- Q.9 Among F and Cl, which one have the higher electron affinity. Give reason for your answer.
- Q.10 Which is more basic, NH₃ of PH₃? Give reason for your answer.

SECTION-C [12 Questions (3 Marks Each), Total Marks: 36]

- Q.11 a) Define Heisenberg Uncertainity Principle?
 - b) Minimum number of electron required to form a bond are?
- Q.12 Give the hybridisation and structure of PCl₃F₂.
- Q.13 The magnetic moment of an element is 3.8 B.M. (Z = 27).
 - a) No. of unpaired electrons are.
 - Write electronic configuration of the element.
- Q.14 A compound contains C = 71.23%, H = 12.95% and O = 15.81%. What is the empirical formula of the compound?
- Q.15 Define Molarity (M), Molality (m) and Normality (N).

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web it is necessary to unravel what its author or authors have said how it has been said when why and where always relating it to its historical moment in this novel part of the article i will explain what changes the irish novelist bram stoker made with respect to the figure of dracula and in the case of mina murray s best friend the wealthy and later vampire lucy