

CHEMISTRY

2019

Time: 20 Minutes

Max. Marks: 17

SECTION A (MULTIPLE CHOICE QUESTIONS)

1. Choose the correct answer for each from the given options.

- (i) The number has five significant figures:
• 391.10 • 300002 • 30000 • 10300
- (ii) At S.T.P., the volume of 3.01×10^{23} molecules of H_2 gas is
• $22.5 dm^3$ • $22.4 dm^3$ • $11.2 dm^3$ • $2.24 dm^3$
- (iii) An ideal gas obeys gas laws at:
• high temperature • high pressure
• all temperatures & pressure • low temperature
- (iv) The process of direct conversion of solid into vapour is called:
• Evaporation • Condensation • Sublimation
• Neutralization
- (v) Capillary action of liquids is due to:
• Viscosity • Surface tension • Density • Fluidity
- (vi) $(n + l)$ value for Sd orbital is:
• 4 • 5 • 6 • 7
- (vii) The colour has the shortest wavelength in the visible spectrum of light:
• Red • Violet • Green • yellow
- (viii) This series of lines is produced when the electron jumps from the 4th orbit to the 2nd orbit:
• Lyman • Balmer • Brackett • Paschen
- (ix) The maximum number of electron can be accommodated in f orbital is: • 7 • 10 • 14 • 15
- (x) This Hydrogen halide has the highest percentage of ionic character: • HF • HCl • HBr • HI
- (xi) The molecule has zero dipole moment:
• NH_3 • CO_2 • HCl • H_2O
- (xii) 1 Debye is equal to: • $2.25 \times 10^{-18} Cm$ • $3.4 \times 10^{-40} Cm$
• $3.335 \times 10^{-23} Cm$ • $3.335 \times 10^{-30} Cm$
- (xiii) One joule is equal to:
• 0.239 cal • 0.391 cal • 0.398 cal • 4.184 cal
- (xiv) In the following reaction, $K_p > K_c$:
• $H_2 + I_2 \rightleftharpoons 2HI$ • $PCl_5 \rightleftharpoons PCl_3 + Cl_2$
• $2SO_2 + O_2 \rightleftharpoons 2SO_3$ • $2NO_2 \rightleftharpoons N_2O_4$
- (xv) Universal indicator in acid given:
• Green colour • Yellow colour
• Pink colour • Red colour
- (xvi) The oxidation number of Mn in $K_2 MnO_4$ is:
• +2 • +4 • +6 • +7

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Marco Cascella



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