Balancing Chemical Equations – Answer Key

Balance the equations below:

- 1 N₂ + 3 H₂ → 2 NH₃
- 2) 2 KClO₃ → 2 KCl + 3 O₂
- 3) 2 NaCl + 1 F₂ → 2 NaF + 1 Cl₂
- 4) 2 H₂ + 1 O₂ → 2 H₂O
- 5) 1 Pb(OH)₂ + 2 HCl → 2 H₂O + 1 PbCl₂
- 6) 2 AlBr₃ + 3 K₂SO₄ → 6 KBr + 1 Al₂(SO₄)₃
- 7) 1 CH₄ + 2 O₂ → 1 CO₂ + 2 H₂O
- 8) 1 C₁H₈ + 5 O₂ → 3 CO₂ + 4 H₂O
- 9) 2 C₈H₁₈ + 25 O₂ → 16 CO₂ + 18 H₂O
- 10) 1 FeCl₃ + 3 NaOH → 1 Fe(OH)₃ + 3 NaCl
- 11) 4 P + 5 O₂ → 2 P₂O₅
- 12) 2 Na + 2 H₂O → 2 NaOH + 1 H₂
- 13) 2 Ag₂O → 4 Ag + 1 O₂
- 14) 1 S₈ + 12 O₂ → 8 SO₃
- 15) 6 CO₂ + 6 H₂O → 1 C₆H₁₂O₆ + 6 O₂
- 16) 1 K + 1 MgBr → 1 KBr + 1 Mg
- 17) 2 HCl + 1 CaCO₃ → 1 CaCl₂ + 1 H₂O + 1 CO₂
- 18) 1 HNO₃ + 1 NaHCO₃ → 1 NaNO₃ + 1 H₂O + 1 CO₂
- 19) 2 H₂O + 1 O₂ → 2 H₂O₂
- 20) 2 NaBr + 1 CaF₂ → 2 NaF + 1 CaBr₂
- 21) 1 H₂SO₄ + 2 NaNO₂ → 2 HNO₂ + 1 Na₂SO₄

Nature Approach To Chemistry Answer Key

David Baud

Nature Approach To Chemistry Answer Key:

This book delves into Nature Approach To Chemistry Answer Key. Nature Approach To Chemistry Answer Key is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Nature Approach To Chemistry Answer Key, encompassing both the fundamentals and more intricate discussions.

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