
Miller & Levine Biology Chapter 2

Questions and Answers

atoms - ANS-the building blocks of matter

proton - ANS-positively charged part of an atom in the nucleus

neutron - ANS-part of atom with no charge in the nucleus

electron - ANS-negatively charged particle of an atom that orbits around the nucleus and is 1/1840 the size of a proton

element - ANS-a pure substance with only one type of atom

compound - ANS-two or more elements chemically combined

ionic bonds - ANS-when one or more electrons are transferred from one atom to the other

covalent bonds - ANS-when electrons are shared between two atoms

molecule - ANS-smallest unit of most compounds

polar - ANS-one end is positive, the other is negative

hydrogen bonds - ANS-hold water molecules, not as strong as chemical bonds but strongest a molecule can form

cohesion - ANS-attraction between molecules of the same substance

adhesion - ANS-attraction between molecules of different substances

mixture - ANS-2 or more elements or compounds physically mixed but not chemically combined and it can be separated

solution - ANS-water and a dissolved substance

solute - ANS-what is being dissolved

solvent - ANS-what it is being dissolved in

suspension - ANS-water and a non-dissolved material

pH scale - ANS-scale that describes how acidic or basic a substance is

Miller And Levine Biology Ch 9 Answers

George Karleskint



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Biology Kenneth Raymond Miller, Prentice Hall (School Division), 1999-02 **Biology** Joseph S. Levine, Kenneth Raymond Miller, 1992 The diversity of life **Prentice Hall Miller Levine Biology Laboratory Manual a for Students Second Edition 2004** Kenneth Raymond Miller, Joseph S. Levine, Prentice-Hall Staff, 2003-02 Authors Kenneth Miller and Joseph Levine continue to set the standard for clear accessible writing and up to date content that engages student interest Prentice Hall Biology utilizes a student friendly approach that provides a powerful framework for connecting the key concepts a biology Students explore concepts through engaging narrative frequent use of analogies familiar examples and clear and instructional graphics Whether using the text alone or in tandem with exceptional ancillaries and technology teachers can meet the needs of every student at every learning level **Prepared to Answer** Rob van de Weghe, 2008

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