

Question			Answer	Marks	Guidance
3	(d)	(ii)	the intermolecular forces are weaker in PH_3 ✓	1	<p>ALLOW the energy needed to overcome the intermolecular forces in NH_3 is greater</p> <p>Check table in part (i)</p> <p>IF NH_3 = hydrogen bonds AND PH_3 = permanent dipoles OR van der Waal's forces; ALLOW 'Hydrogen Bonds are stronger' ORA</p> <p>IF NH_3 = permanent dipoles AND PH_3 = van der Waal's forces; ALLOW 'permanent dipoles are stronger' ORA</p> <p>IF NH_3 = permanent dipoles AND PH_3 = permanent dipoles; ALLOW 'permanent dipoles are stronger in NH_3' ORA</p> <p>DO NOT ALLOW PH_3 has weaker vdW's than NH_3 DO NOT ALLOW NH_3 has stronger hydrogen bonds than PH_3 DO NOT ALLOW implication that covalent bonds are broken</p>
	(e)	(i)	Both electrons have been donated by one atom ✓	1	<p>ALLOW 'they' for electrons IGNORE elements for atom DO NOT ALLOW 'transfer' in place of 'donated' DO NOT ALLOW more than one electron pair is donated</p>

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Table of Contents Jun 2013 Ocr F58paper

1. Understanding the eBook Jun 2013 Ocr F58paper
 - The Rise of Digital Reading Jun 2013 Ocr F58paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Jun 2013 Ocr F58paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jun 2013 Ocr F58paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jun 2013 Ocr F58paper
 - Personalized Recommendations
 - Jun 2013 Ocr F58paper User Reviews and Ratings
 - Jun 2013 Ocr F58paper and Bestseller Lists

5. Accessing Jun 2013 Ocr F58paper Free and Paid eBooks
 - Jun 2013 Ocr F58paper Public Domain eBooks
 - Jun 2013 Ocr F58paper eBook Subscription Services
 - Jun 2013 Ocr F58paper Budget-Friendly Options
6. Navigating Jun 2013 Ocr F58paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Jun 2013 Ocr F58paper Compatibility with Devices
 - Jun 2013 Ocr F58paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jun 2013 Ocr F58paper
 - Highlighting and Note-Taking Jun 2013 Ocr F58paper
 - Interactive Elements Jun 2013 Ocr F58paper
8. Staying Engaged with Jun 2013 Ocr F58paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jun 2013 Ocr F58paper
9. Balancing eBooks and Physical Books Jun 2013 Ocr F58paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jun 2013 Ocr F58paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Jun 2013 Ocr F58paper
 - Setting Reading Goals Jun 2013 Ocr F58paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Jun 2013 Ocr F58paper
 - Fact-Checking eBook Content of Jun 2013 Ocr F58paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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