Questions courtes

Vrai/Faux

Corrigé p. 449

Toutes les variables aléatoires de cet exercice sont définies sur un même univers Ω .

1. Si X et Y sont deux variables de Bernoulli telles que :

 $p([X = 1] \cap [Y = 1]) = p(X = 1) \times p(Y = 1)$, alors X et Y sont indépendantes.

2. Si X et Y sont deux variables aléatoires à valeurs dans {1, 2, ..., n}, alors on a :

$$\sum_{j=1}^{n} \left[\sum_{i=1}^{n} p([X=i] \cap [Y=j]) \right] = 1.$$

3. Soit X et Y deux variables aléatoires à valeurs dans $\{1, 2, ..., n\}$, et i un entier tel que $p(X = i) \neq 0$. Alors on a :

$$\forall j \in \{1, 2, ..., n\}, \quad p(Y = j) = \sum_{i=1}^{n} p_{[X = i]}(Y = j).$$

4. Si X et Y ont la même loi, alors elles sont indépendantes.

5.Si p(X = i) ≠ 0 alors on a:

$$\sum_{j \in Y(\Omega)} P_{[X=i]}(Y=j) = 1.$$

6. Si X et Y sont deux variables indépendantes suivant la même loi de Bernoulli de paramètre p, alors X + Y suit une loi usuelle.

7. Si X_1 et X_2 sont deux variables indépendantes telles que X_1 suit la loi binomiale $\mathfrak{B}(n,p_1)$ et X_2 suit la loi binomiale $\mathfrak{B}(n,p_2)$ alors X_1+X_2 suit la loi binomiale $\mathfrak{B}(2n,p_1+p_2)$.

8. Si M_n est la moyenne d'un échantillon $(X_1, X_2, ..., X_n)$ d'une loi d'espérance m et d'écart-type σ , alors :

$$M_n = \frac{X_1 + X_2 + ... + X_n}{n}$$
, $E(M_n) = m$ et $V(M_n) = \frac{\sigma^2}{n}$.

9. Si la loi de X est donnée par $p(X = 1) = p(X = -1) = \frac{1}{2}$ et si Y est indépendante de X et suit la même loi que X, alors X + Y prend les valeurs -2, 0 et 2 avec équiprobabilité.

10. Si X et Y sont deux variables de Bernoulli de paramètre p(0

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web what is an oxidation reduction or redox reaction this is a reaction in which electrons are transferred between reactants what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is a helpful way to remember this

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a i2o5 5co i2 5co i 2 5 c o i 2 5 c o i 2 5 c o i

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons chapter 20 worksheet redox beverly hills high school - Jul 02 2022

web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also 2sr o2 2sro sr0 to sr2 oxidized reducing agent o0 to o2 reduced ox ag 2li s li2s li0 to li1 oxidized red ag s0 to s2 reduced ox ag 3 2cs br2