

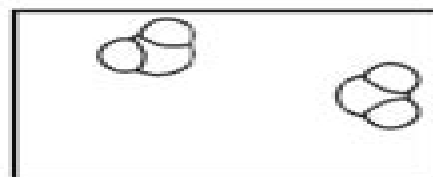
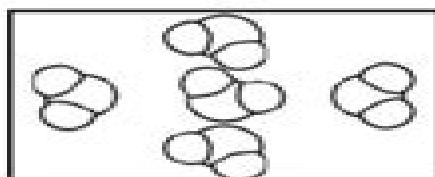
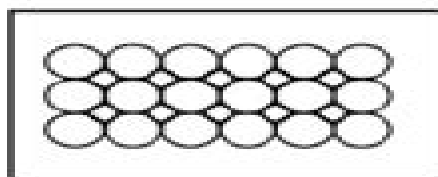
## States of Matter and Phase Changes

### States of Matter

All forms of matter can exist in one of the four states of matter:

- **Solids** have a definite shape and volume. Atoms or molecules in a solid are held firmly in place, but can still vibrate. Solids have the least amount of energy.
- **Liquids** have a definite volume, but take the shape of their container. The atoms or molecules of a liquid are able to flow around each other.
- **Gases** contain molecules that fly at great speeds. Gases also take both the shape and volume of their container, assuming they are in one.
- **Plasma** is the highest-energy state of matter. In plasmas, the particles that make up the atoms and molecules escape and can move freely around.

The three diagrams below show the states of water. Label each as either a gas, liquid, or solid.

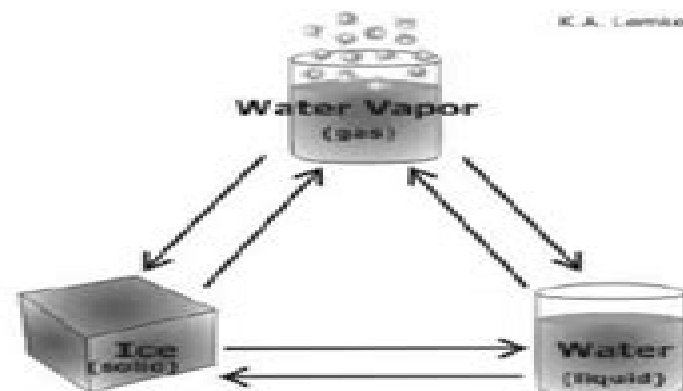


### Phase Changes

Matter is most often converted from one state to another by adding or removing heat.

- **Vaporization** converts from liquid to gas.
- **Condensation** converts from gas to liquid.
- **Freezing** converts from liquid to solid.
- **Melting** converts from solid to liquid.
- **Deposition** converts from gas to solid.
- **Sublimation** converts from solid to gas.

Label each phase change on the diagram to the right.



# Matter And Change Practice Problems Answers

**Kenneth W. Raymond**



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