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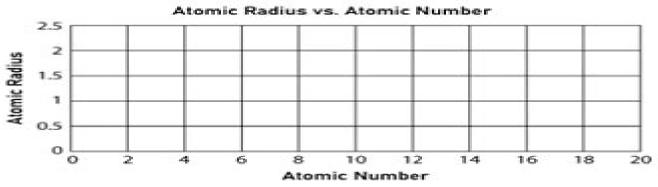
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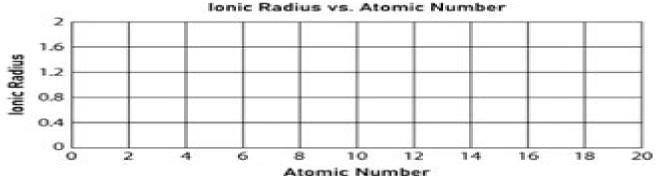
The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

Element	H	He	Li	Be	В	C	N	0	F
Atomic Number	1	2	3	4	5	6	7	8	9
Atomic Radius (Å)	0.79	0.49	2.05	1.4	1.17	0.91	0.75	0.65	0.57
lonic Radius (Å)	0.012	= - 3:	0.76	0.35	0.23	0.16	0.13	1.4	1.33

Ne	Na	Mg	Al	Si	P	S	CI	Ar
10	1.1	12	13	14	15	16	17	18
0.38	2.23	1.72	1.82	1.46	1.23	1.09	0.97	0.71
	1.02	0.72	0.535	0.4	0.38	0.37	1.81	-

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.





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Robert C. Smoot, Smoot, Richard G. Smith, Jack Price

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about the outcome of the experiment b To formulate a hypothesis c To think about a phenomenon in a specific way d To apply personal experience in answering a questions Pref Prentice Hall Science Explorer Michael J. Padilla,2002 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 The Software Encyclopedia 2001, 2001

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Table of Contents Lab Periodic Trends Key

- 1. Understanding the eBook Lab Periodic Trends Key
 - The Rise of Digital Reading Lab Periodic Trends Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Lab Periodic Trends Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lab Periodic Trends Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Lab Periodic Trends Key
 - Personalized Recommendations
 - Lab Periodic Trends Key User Reviews and Ratings
 - Lab Periodic Trends Key and Bestseller Lists
- 5. Accessing Lab Periodic Trends Key Free and Paid eBooks
 - Lab Periodic Trends Key Public Domain eBooks
 - Lab Periodic Trends Key eBook Subscription Services

- Lab Periodic Trends Key Budget-Friendly Options
- 6. Navigating Lab Periodic Trends Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Lab Periodic Trends Key Compatibility with Devices
 - Lab Periodic Trends Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lab Periodic Trends Key
 - Highlighting and Note-Taking Lab Periodic Trends Key
 - Interactive Elements Lab Periodic Trends Key
- 8. Staying Engaged with Lab Periodic Trends Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lab Periodic Trends Key
- 9. Balancing eBooks and Physical Books Lab Periodic Trends Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lab Periodic Trends Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Lab Periodic Trends Key
 - Setting Reading Goals Lab Periodic Trends Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Lab Periodic Trends Key
 - Fact-Checking eBook Content of Lab Periodic Trends Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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