

Mole Ratios POGIL Part 2

What mass of nitrogen is needed to produce 30.0 g of ammonia?

we have: 30.0 g of ammonia (NH_3)

we want: grams of nitrogen (N_2)



Step 1: get moles! (use molar mass)

$$\frac{30.0 \text{ g NH}_3}{17.04 \text{ g NH}_3} \times 1 \text{ mol NH}_3 = 1.76 \text{ mol NH}_3$$



More Ratios Pogil Answer

Sunil Tanna



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Mole Ratios Lifeliqe,2019 This lesson plan relates balanced chemical equations to everyday analogies defines stoichiometry and uses mole ratios to convert between amounts of substances in chemical reactions **STOICHIOMETRY** NARAYAN CHANGDER,2024-04-01 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today's academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams quizzes trivia and more

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