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## Physical Setting Chemistry Regents January 2025 Questions And Answers 2025 Actual Exam!!

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At standard pressure, compare the strength of IMF in  $\text{NO}_2$  (g) and  $\text{N}_2$  (g) -  
At STP, \* $\text{NO}_2$  has stronger IMF than  $\text{N}_2$ \*.

Based on Table F, which equation represents a saturated solution having  
the lowest concentration of  $\text{Cl}^-$  ions?

a)  $\text{NaCl} \rightarrow \text{Na}^+ + \text{Cl}^-$

# January 24 2013 Chemistry Regents Answers

**Sharon H. Welcher**



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