

**CLASSIQUES
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COLLÈGE**

Michel Onfray

Le Corps de mon père

Autobiographie de ma mère

TEXTE INTÉGRAL



Le Tennis Ma Sauveacute Autobiographie

TD Snyder



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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq}$ $\text{Cr}_2\text{O}_7^{2-} \text{aq}$

rightarrow aq 2cr 2o 7s label 4 2 3

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is 2na aq 2cl aq cu 2aq so 42aq 2na aq so 42aq cucl 2s the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web the six no 3aq ions and the six na aq ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation 3ba 2aq 2po 4aq 3aq \rightarrow ba 3po 4aq 2s

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water h aq oh aq h

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be

2cl aq cu 2 aq cucl 2 s 2cl aq cu 2 aq cucl 2 s these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length

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