

Unit 6 – Worksheet 3

Ionic Compounds

Properties

Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- a. Na_2O
- b. K_2S
- c. MgCl_2
- d. CaBr_2
- e. BaI_2
- f. Al_2S_3
- g. CsBr
- h. AgF

2. Give the name of the following simple binary ionic compounds.

- a. Na_3N
- b. K_2O
- c. AgBr
- d. MgI_2
- e. SrO

3. Write the formula for the following binary ionic compounds.

- a. lithium bromide
- b. sodium iodide
- c. silver sulfide
- d. cesium oxide
- d. beryllium iodide
- f. barium hydride
- g. aluminum fluoride
- h. potassium oxide

4. Write the formula for these ionic substances.

- a. silver oxide
- b. aluminum sulfide
- c. sodium nitride
- d. barium chloride
- e. strontium hydride

Modeling Chemistry U6 Test

Yi-Tong Ma



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Computational Methods in Lanthanide and Actinide Chemistry Michael Dolg, 2015-02-17 The f elements and their compounds often possess an unusually complex electronic structure governed by the high number of electronic states arising from open f shells as well as large relativistic and electron correlation effects A correct theoretical description of these elements poses the highest challenges to theory Computational Methods in Lanthanide and Actinide Chemistry summarizes state of the art electronic structure methods applicable for quantum chemical calculations of lanthanide and actinide systems and presents a broad overview of their most recent applications to atoms molecules and solids The book contains sixteen chapters written by leading experts in method development as well as in theoretical investigations of f element systems Topics covered include Relativistic configuration interaction calculations for lanthanide and actinide anions Study of actinides by relativistic coupled cluster methods Relativistic all electron approaches to the study of f element chemistry Relativistic pseudopotentials and their applications Gaussian basis sets for lanthanide and actinide elements Applied computational actinide chemistry This book will serve as a comprehensive reference work for quantum chemists and computational chemists both those already working in and those planning to enter the field of quantum chemistry for f elements Experimentalists will also find important information concerning the capabilities of modern quantum chemical methods to assist in the interpretation or even to predict the outcome of their experiments **The Physics and Chemistry**

of Solids, 1958 *29th European Symposium on Computer Aided Chemical Engineering* Anton A. Kiss, Edwin Zondervan, Richard Lakerveld, Leyla Özkan, 2019-06-28 The 29th European Symposium on Computer Aided Process Engineering contains the papers presented at the 29th European Symposium of Computer Aided Process Engineering ESCAPE event held in Eindhoven The Netherlands from June 16 19 2019 It is a valuable resource for chemical engineers chemical process engineers researchers in industry and academia students and consultants for chemical industries Presents findings and discussions from the 29th European Symposium of Computer Aided Process Engineering ESCAPE event

Index Medicus, 2001 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings **The Journal of Physics and Chemistry of Solids**, 1958 *Energy Research Abstracts*, 1986 **Progress Report for Period Ending Mar. 31, 1950-**. Oak Ridge National Laboratory. Analytical Chemistry Division, 1965 **Mathematics, Physics, Chemistry, and Engineering**, 1996 **Scientific and Technical Aerospace Reports**, 1986 **Corrosion** Stephen D. Cramer, Bernard S. Covino, 2003 The first book in a two volume revision of the 1987 Metals Handbook 9th edition addresses the needs of the global technical community for current information Chapters on fundamentals cover the theory of aqueous and gaseous corrosion from thermodynamic and kinetic perspectives while chapters on forms of corrosion tell how to recognize different types and the forces that influence them Testing and evaluation methods are addressed as are methods of protection and topics related to redesigning for corrosion control and prevention A section on tools for the corrosionist provides conventions

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