

Unit 7: Balancing Chemical Reactions- Worksheet 2

Balance the following equations by inserting the proper coefficients.

1. $\underline{2}\text{SO}_2 + \underline{1}\text{O}_2 \rightarrow \underline{2}\text{SO}_3$ **Synthesis**
 2. $\underline{2}\text{CH}_4 + \underline{3}\text{O}_2 \rightarrow \underline{2}\text{CO} + \underline{4}\text{H}_2\text{O}$ **Single R**
 3. $\underline{2}\text{P} + \underline{3}\text{Cl}_2 \rightarrow \underline{2}\text{PCl}_3$ **Synthesis**
 4. $\underline{2}\text{CO} + \underline{1}\text{O}_2 \rightarrow \underline{2}\text{CO}_2$ **Synthesis**
 5. $\underline{2}\text{CH}_4 + \underline{1}\text{O}_2 \rightarrow \underline{2}\text{CH}_3\text{OH}$ **Synthesis**
 6. $\underline{2}\text{Li} + \underline{1}\text{Br}_2 \rightarrow \underline{2}\text{LiBr}$ **Synthesis**
 7. $\underline{2}\text{Al}_2\text{O}_3 \rightarrow \underline{4}\text{Al} + \underline{3}\text{O}_2$ **Decomp**
 8. $\underline{2}\text{Na} + \underline{2}\text{H}_2\text{O} \rightarrow \underline{2}\text{NaOH} + \underline{1}\text{H}_2$ **Single R**
 9. $\underline{6}\text{CO}_2 + \underline{6}\text{H}_2\text{O} \rightarrow \underline{1}\text{C}_6\text{H}_{12}\text{O}_6 + \underline{6}\text{O}_2$ **Rev. Comb**
 10. $\underline{1}\text{H}_2\text{SO}_4 + \underline{2}\text{NaCl} \rightarrow \underline{2}\text{HCl} + \underline{1}\text{Na}_2\text{SO}_4$ **Double R**
- Two reactions used to get rid of sulfur dioxide, a pollutant from burning coal:
11. $\underline{3}\text{H}_2 + \underline{1}\text{SO}_2 \rightarrow \underline{1}\text{H}_2\text{S} + \underline{2}\text{H}_2\text{O}$ **Double R**
 12. $\underline{2}\text{CaCO}_3 + \underline{2}\text{SO}_2 + \underline{1}\text{O}_2 \rightarrow \underline{2}\text{CaSO}_4 + \underline{2}\text{CO}_2$ **No**
 13. $\underline{2}\text{AgNO}_3 + \underline{1}\text{CaCl}_2 \rightarrow \underline{2}\text{AgCl} + \underline{1}\text{Ca}(\text{NO}_3)_2$ **Double R**
 14. $\underline{2}\text{HCl} + \underline{1}\text{Ba}(\text{OH})_2 \rightarrow \underline{1}\text{BaCl}_2 + \underline{2}\text{H}_2\text{O}$ **Double R**
 15. $\underline{1}\text{H}_3\text{PO}_4 + \underline{3}\text{NaOH} \rightarrow \underline{1}\text{Na}_3\text{PO}_4 + \underline{3}\text{H}_2\text{O}$ **Double R**
 16. $\underline{1}\text{Pb}(\text{NO}_3)_2 + \underline{2}\text{KI} \rightarrow \underline{1}\text{PbI}_2 + \underline{2}\text{KNO}_3$ **Double R**
 17. $\underline{3}\text{CuO} + \underline{2}\text{NH}_3 \rightarrow \underline{1}\text{N}_2 + \underline{3}\text{Cu} + \underline{3}\text{H}_2\text{O}$ **Single R**
 18. $\underline{1}\text{C}_2\text{H}_5\text{OH} + \underline{3}\text{O}_2 \rightarrow \underline{2}\text{CO}_2 + \underline{3}\text{H}_2\text{O}$ **Com**
 19. $\underline{2}\text{C}_2\text{H}_6 + \underline{3}\text{O}_2 \rightarrow \underline{2}\text{CH}_3\text{COOH} + \underline{2}\text{H}_2\text{O}$ **No**
 20. $\underline{3}\text{NO}_2 + \underline{1}\text{H}_2\text{O} \rightarrow \underline{2}\text{HNO}_3 + \underline{1}\text{NO}$ **No**

Modeling Chemistry Ws Answers Balancing Chemical Reaction

RS Peters



Modeling Chemistry Ws Answers Balancing Chemical Reaction:

Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 **Handbook of Formative Assessment** Heidi Andrade, Gregory J. Cizek, 2010-04-02

Formative assessment has recently become a focus of renewed research as state and federal policy makers realize that summative assessments have reached a point of diminishing returns as a tool for increasing student achievement. Consequently, supporters of large scale testing programs are now beginning to consider the potential of formative assessments to improve student achievement. The mission of this handbook is to comprehensively profile this burgeoning field of study. Written by leading international scholars and practitioners, each chapter includes a discussion of key issues that dominate formative assessment policy and practice today as well as those that are likely to affect research and practice in the coming years. Key features include:

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Balancing Chemical Equations Worksheet
Crispin Collins, 2020-09-12

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Teacher Instruction and Student Performance in Balancing Chemical Equations
William Levi Yaroch, 1983

Euro-Par 2012: Parallel Processing Workshops Ioannis Caragiannis, Michael Alexander, Rosa Maria Badia, Mario Cannataro, Alexandru Costan, Marco Danelutto, Frédéric Desprez, Bettina Krammer, Julio Sahuquillo, Stephen L. Scott, Josef Weidendorfer, 2013-02-15

This book constitutes thoroughly refereed post conference proceedings of the workshops of the 18th International Conference on Parallel Computing Euro Par 2012 held in Rhodes.

Islands Greece in August 2012 The papers of these 10 workshops BDMC CGWS HeteroPar HiBB OMHI Paraphrase PROPER UCHPC VHPC focus on promotion and advancement of all aspects of parallel and distributed computing

Physical Science Carson-Dellosa Publishing, 2015-03-09 Physical Science for grades 5 to 12 is designed to aid in the review and practice of physical science topics Physical Science covers topics such as scientific measurement force and energy matter atoms and elements magnetism and electricity The book includes realistic diagrams and engaging activities to support practice in all areas of physical science The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series is aligned to current science standards Holt Chemistry Ralph Thomas Myers, 2004 *EPA Publications Bibliography* United States. Environmental Protection Agency, 1997 *EPA Publications Bibliography Quarterly Abstract Bulletin* United States. Environmental Protection Agency, 1997-10

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

Introduction to General, Organic, and Biochemistry Morris Hein, 1997 The authors recognize that both science and mathematics may be daunting subjects for many students taking this course With this in mind they have anticipated where students might stumble and have paced and organized this text to help them through Their goal is to make the material interesting and relevant so students understand the basic chemical principles related to their career The authors emphasize problem solving and provide a range of practice exercises As in previous editions the text first presents the basic concepts of general chemistry and then moves into organic and biochemistry In this edition the first two sections have been revised primarily to improve explanations and include new pedagogical features The biochemistry portion has been thoroughly updated to include coverage of many recent developments and emerging technologies in the field

Current Index to Journals in Education, 1989 *Balancing Chemical Equations Worksheets (Over 200 Reactions to Balance)* Chris McMullen, 2016-01-12 Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas Not just for students Anyone who enjoys math and science puzzles can enjoy the challenge of balancing these

chemical reactions Illinois Chemistry Teacher ,1992 Christian Home Educators' Curriculum Manual Cathy Duffy,1997-11 **The Science Teacher** ,1992 Some issues are accompanied by a CD ROM on a selected topic *The Software Encyclopedia 2001* ,2001 **Agronomy Abstracts** ,1987 Includes abstracts of the annual meetings of the American Society of Agronomy Soil Science Society of America Crop Science Society of America of its Agronomic Education Division *English Mechanic and Mirror of Science and Art* ,1891 English Mechanic and Mirror of Science ,1886

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